

In situ and Operando XAS characterization of Functional Materials

Anthony Beauvois



*anthony.beauvois@synchrotron-soleil.fr

Ex Situ

Determination of functional **materials activity at the laboratory**
(*catalysis, battery cycling,...*)

Characterization done separately
(*Fresh and post-mortem sample*)

Multimodal characterization

UV-vis, Raman, XRD,...



Preparation of optimized samples (e.g. pellets)



Modification of sample after activity (re-oxidation,...)?
What happen *during* the reaction?

In Situ

Measurements on materials performed in a cell allowing to apply reaction conditions to the functional materials
(*Heating, gas flowing, potential cycling, pressure,...*)

Operando

Introduced by M.A. Bañares in 2002

Measurements on materials performed in a cell allowing:
real reaction conditions and simultaneously determination of the activity
(*product detection, current production,...*)



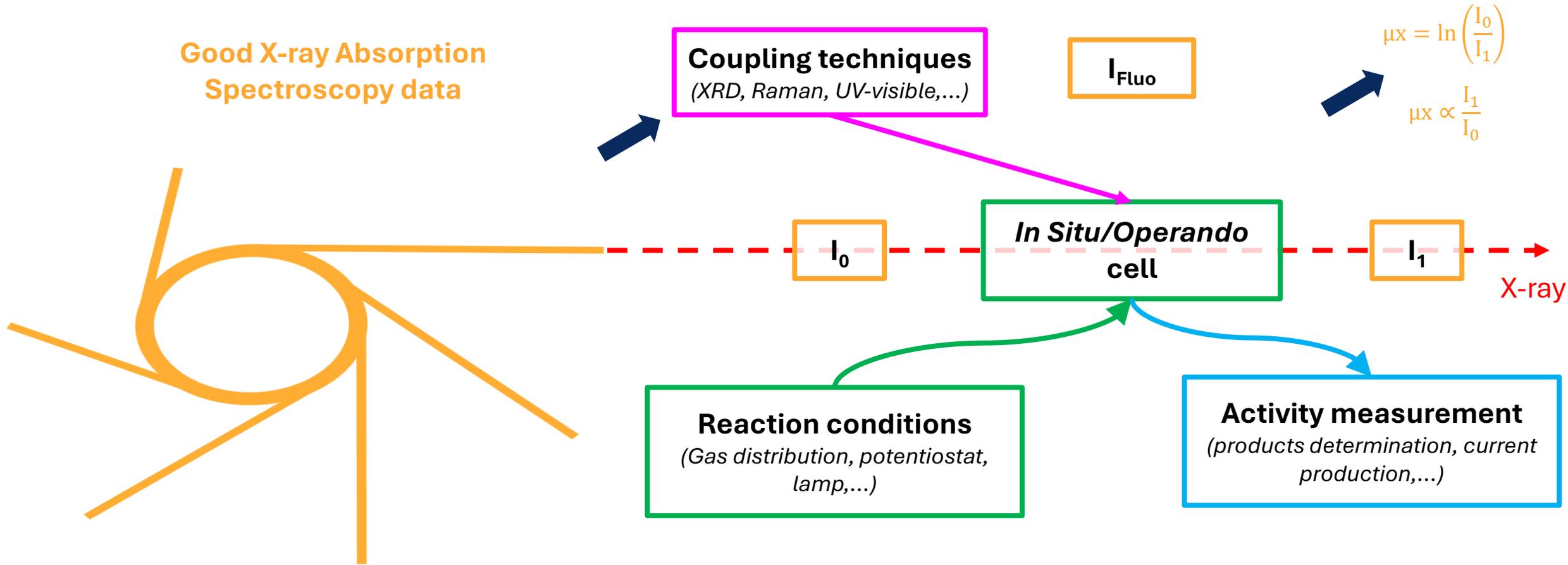
Establishment of structure-activity relationship in real conditions



Are the observed transformations related to changes in the activity?

What is needed for *operando* characterization using XAS?

Good X-ray Absorption Spectroscopy data



Cells able to reproduce real reaction conditions

Qualitative and Quantitative access to the material activity

Complementary Methods for simultaneous Characterization

Large dataset analysis methodology

I. Examples of *in situ/operando* setups

Catalysis (thermal catalysis, (photo)-(electro)-catalysis)

Energy storage

Liquid cells

II. Multimodal characterisations and data analysis methodology

III. Scientific case: LDH used for Ethanol Stream Reforming

IV. Conclusions

What is needed for *operando* characterization using XAS?

Good X-ray Absorption Spectroscopy data

Coupling techniques
(XRD, Raman, UV-visible,...)

I_{Fluo}

$$\mu_x = \ln\left(\frac{I_0}{I_1}\right)$$

$$\mu_x \propto \frac{I_1}{I_0}$$

I_0

In Situ/Operando cell

I_1

X-ray

Reaction conditions
(Gas distribution, potentiostat, lamp,...)

Activity measurement
(products determination, current production,...)

Cells able to reproduce real reaction conditions

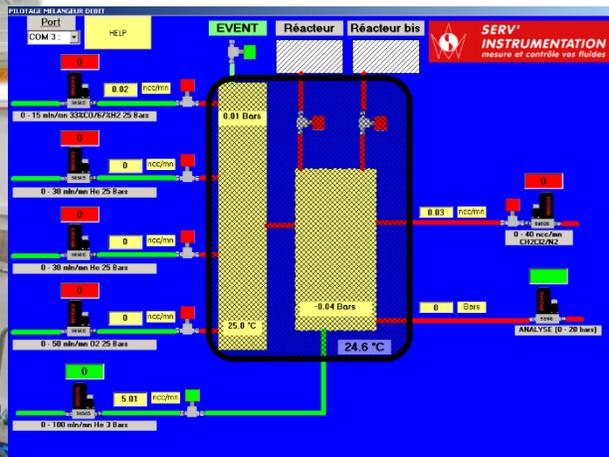
Qualitative and Quantitative access to the material activity

Complementary Methods for simultaneous Characterization

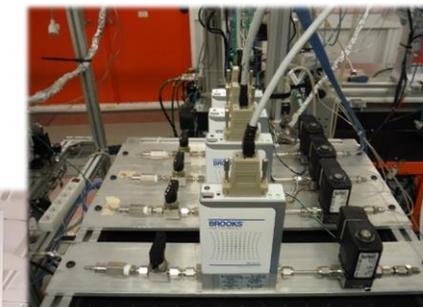
Large dataset analysis methodology

Gas distribution system

Ventilated cabinet for safety
with dangerous gases



- Complex mixtures
- Saturator for liquid reactants
- Heated lines (120°C)
- $P_{atm} \rightarrow 20 \text{ bar}$
- Remote control

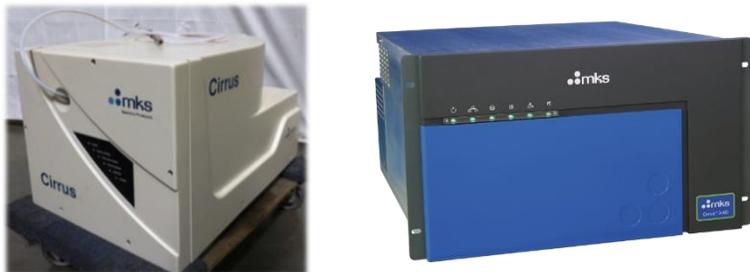


- Saturator for liquid reactants
- Fast switching between gases
- P_{atm}
- Remote control

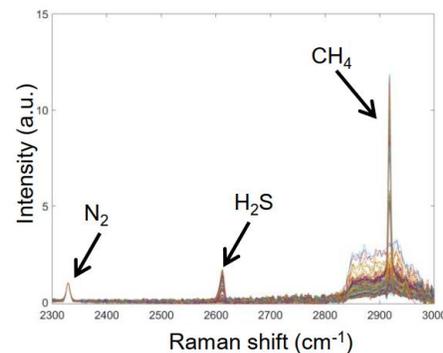
+ Welcome of users'
equipments

Products analysis

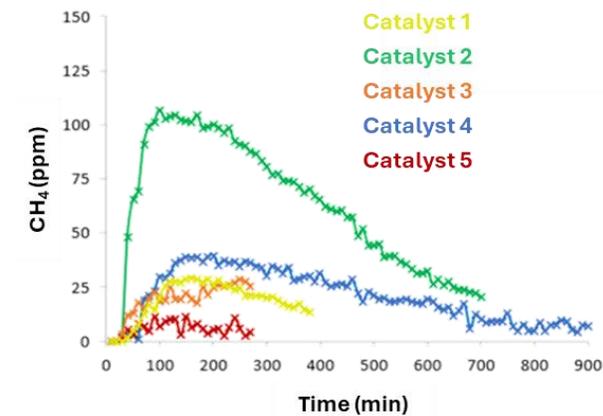
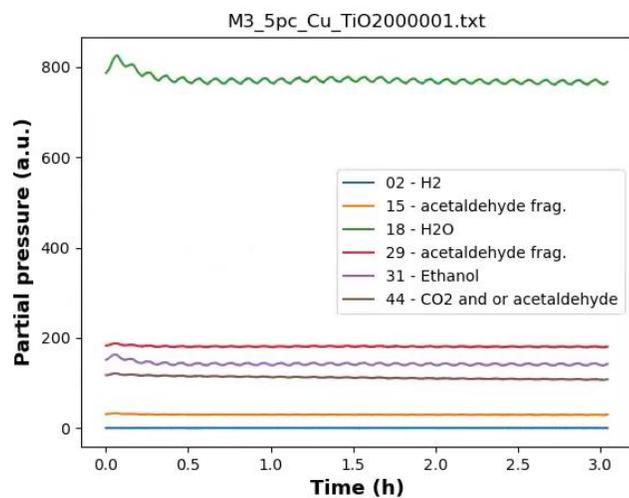
Mass Spectrometer



Raman probe

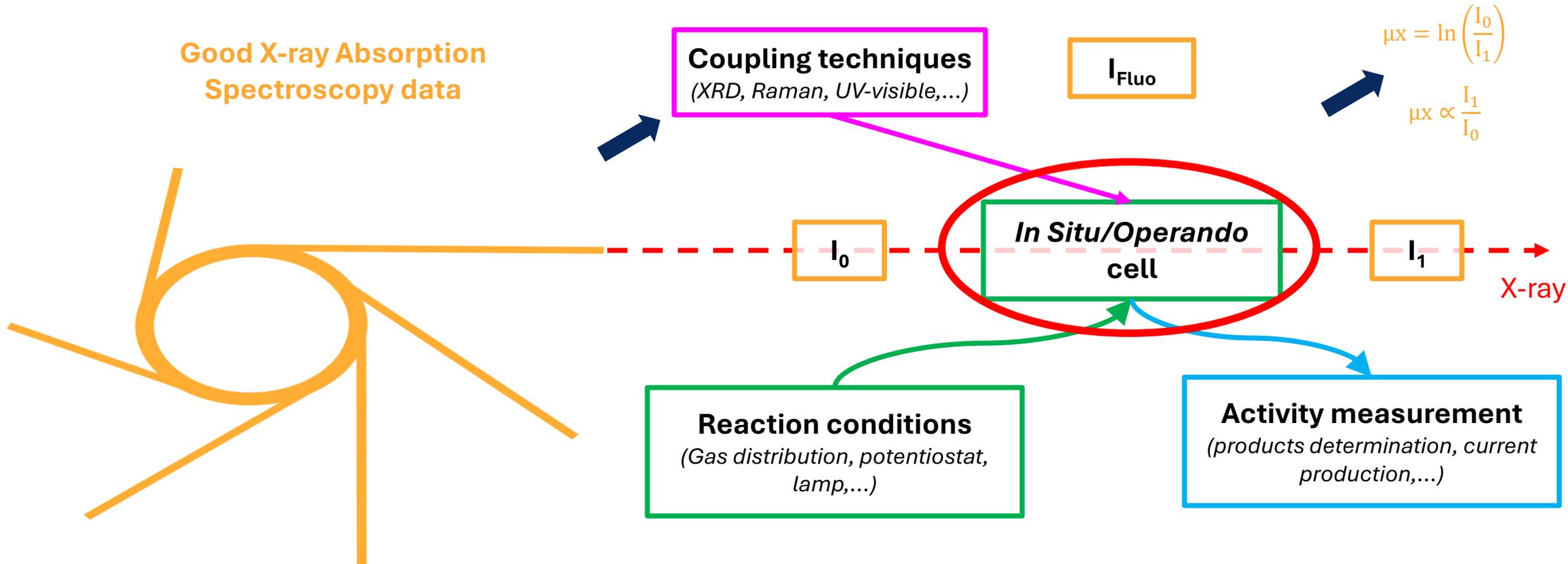


Micro-GC



What is needed for *operando* characterization using XAS?

Good X-ray Absorption Spectroscopy data



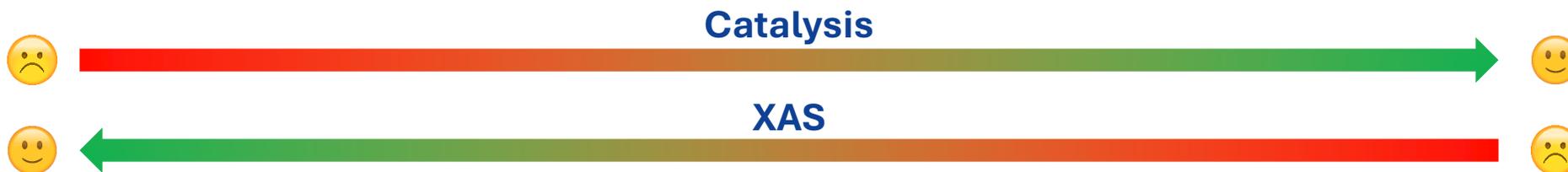
Cells able to reproduce real reaction conditions

Qualitative and Quantitative access to the material activity

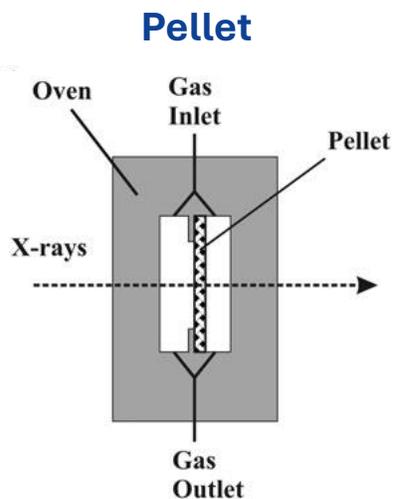
Complementary Methods for simultaneous Characterization

Large dataset analysis methodology

What is a good cells for gas-phase *operando* catalysis?



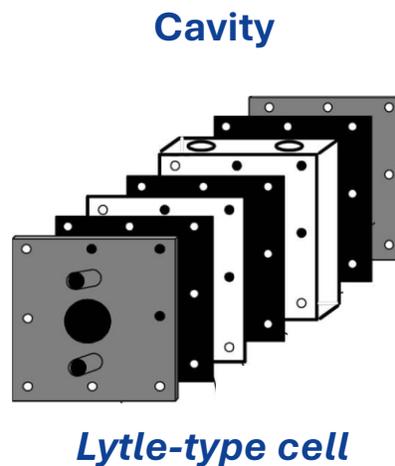
PACKED BED REACTOR



Jentoft *et al.*, *Rev Sci. Ins.*, 1996, **67**, 2111
 Grundwaldt *et al.*, *Phys. Chem. Chem. Phys.*, 2004, **6**, 3037

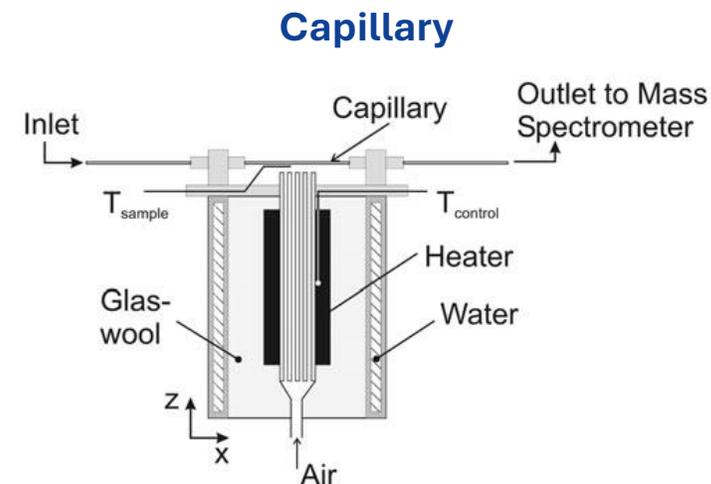
**Bad system
for catalysis**

PLUG-FLOW REACTOR



Lytle *et al.*, *J. Chem. Phys.*, 1979, **70**, 4849
 Ravel *et al.*, *Nim B*, 1999, **183**
 Girardon *et al.*, *J. Synch. Rad.*, 2005, **12**, 680
 Kawai *et al.*, *Rev. Sc. Ins.*, 2008, **79**, 1

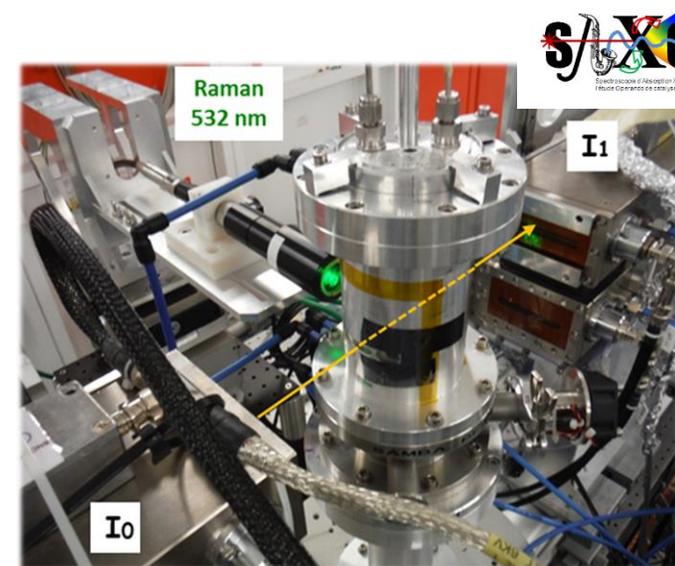
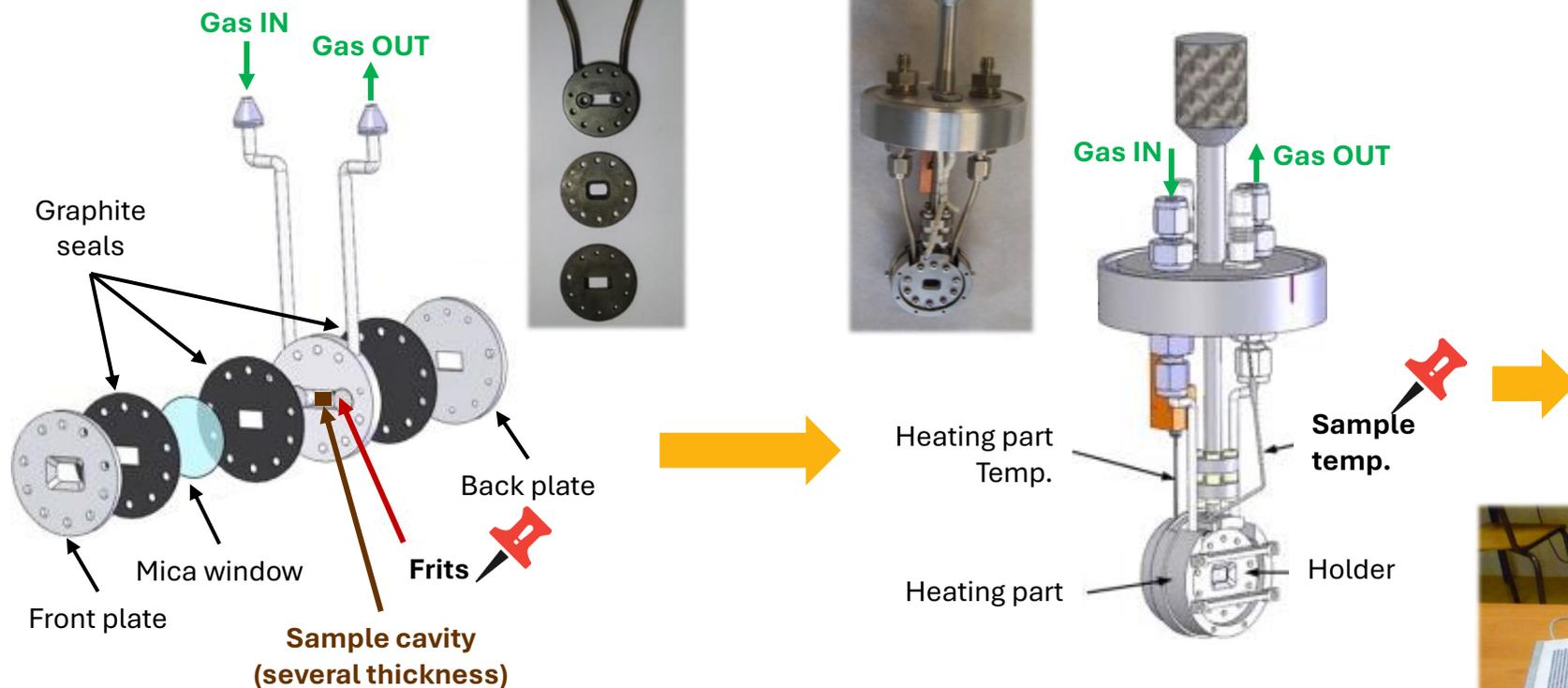
**Good
compromise**



Clausen *et al.*, *J. Cat.*, 1991, **132**, 524
 Grundwaldt *et al.*, *Phys. Chem. Chem. Phys.*, 2004, **6**, 3037

**• Heterogeneity problems for XAS
• Not easy to handle**

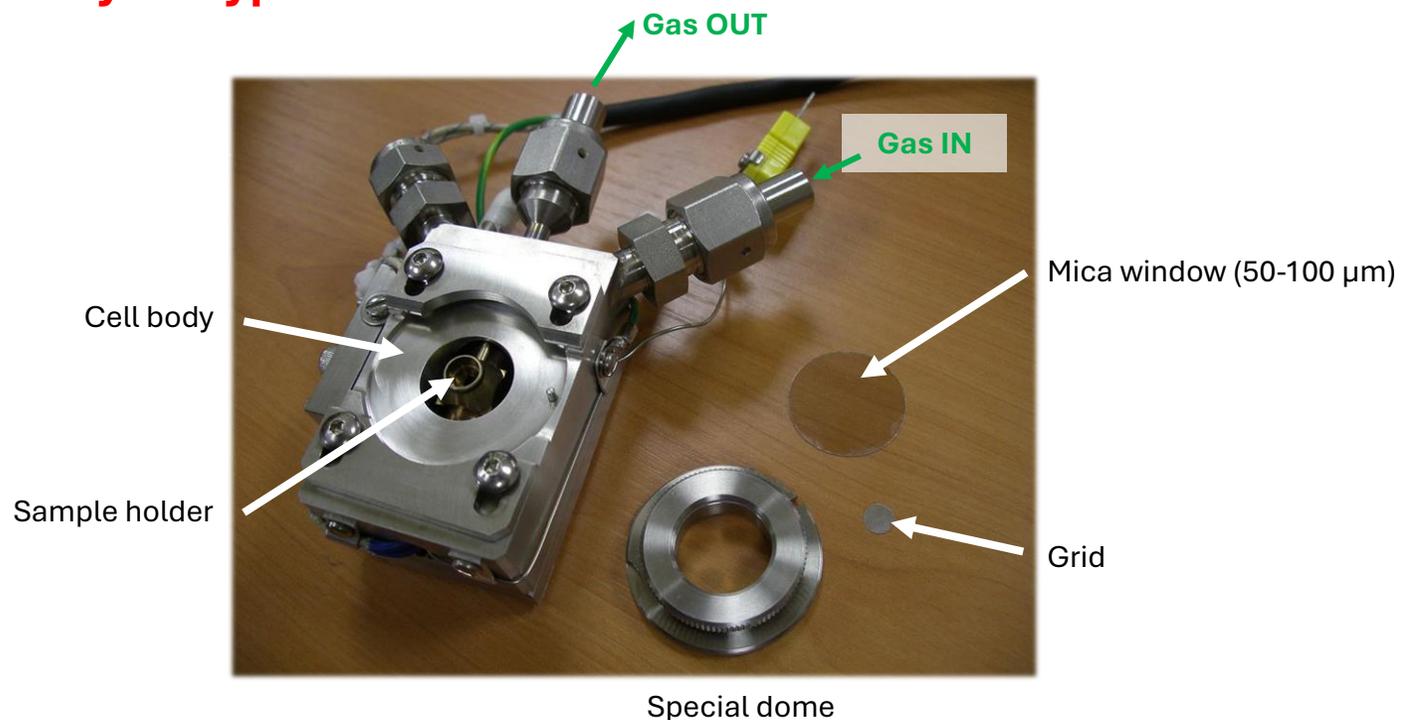
Thermal catalysis Lytle-type cell



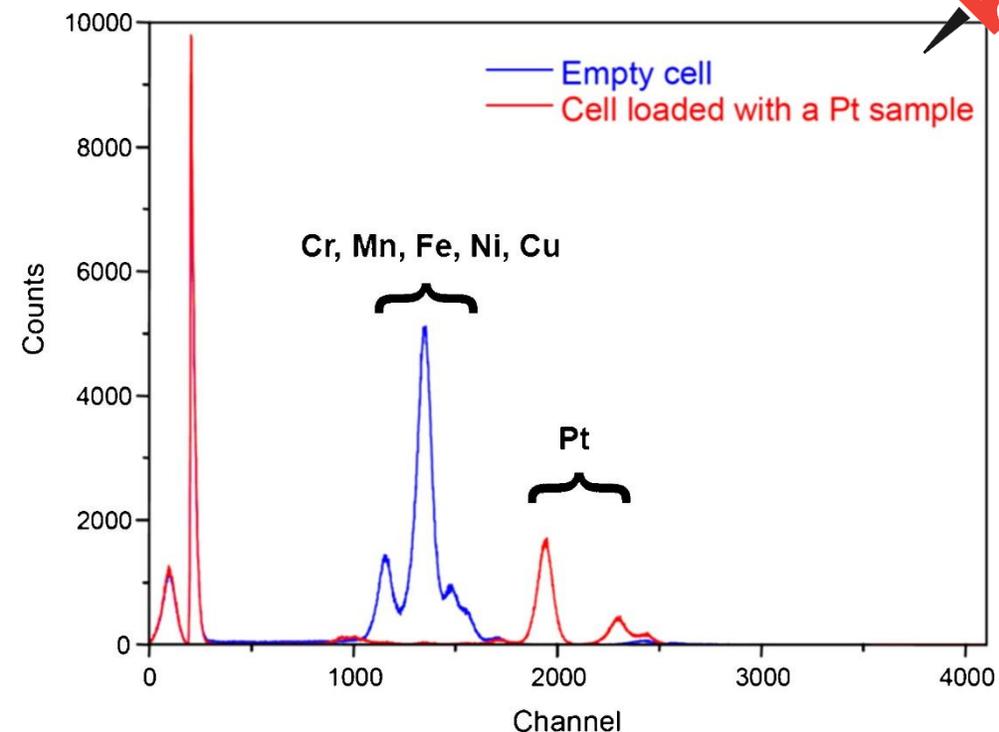
Temperature	Room temp. → 600°C
Pressure	1 bar
Sample	Powdered catalyst gently pressed in the cavity
XAS measure	Fluorescence + Transmission
Multimodal	XRD, Raman

La Fontaine *et al.*, *Cat. Today*, 2013, **205**, 148-158

Thermal catalysis Lytle-type cell



Chemical composition of the cell using X-ray fluorescence

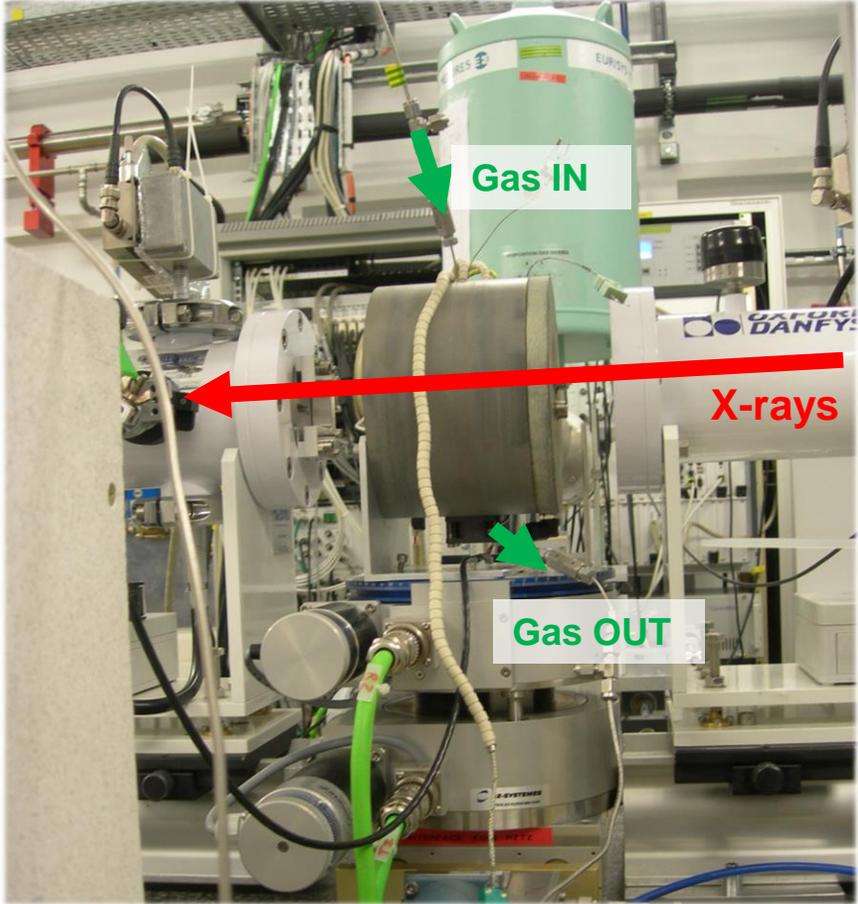
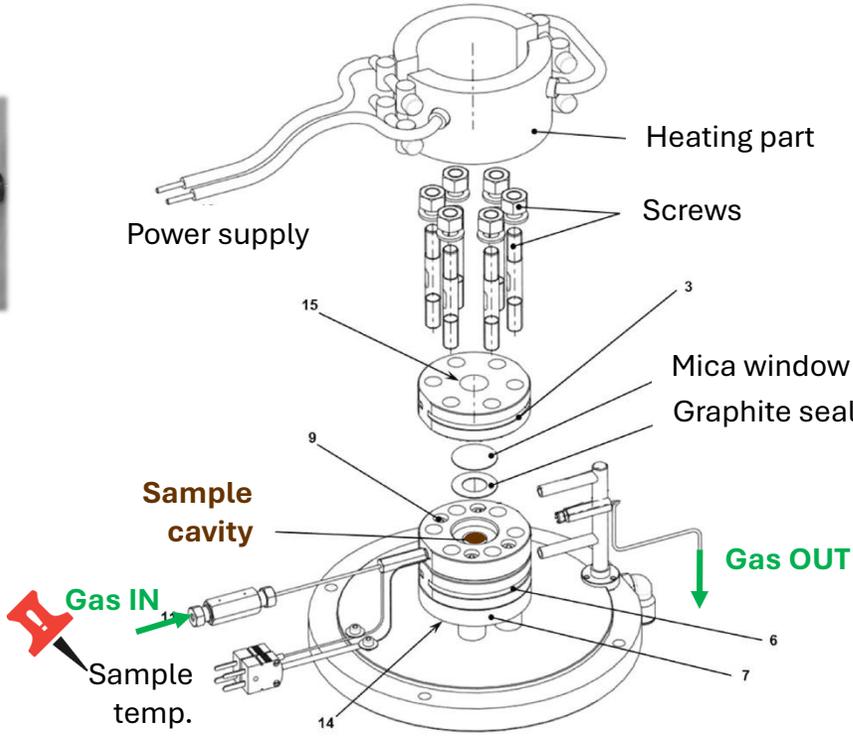
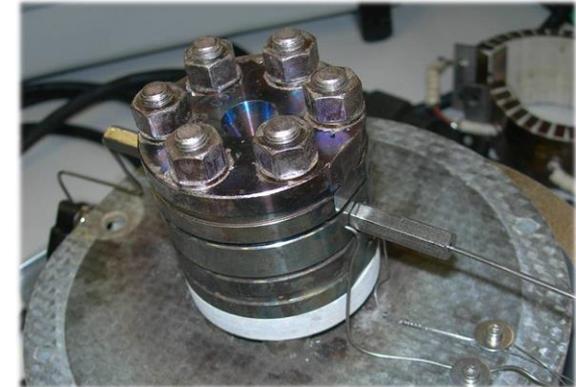


- Take care of the cell/sample holder composition, especially for fluorescence
- Effect reduced by:
 - Geometry of the cell (windows,...)
 - Presence of the sample

Temperature	Room temp. → 500°C
Pressure	1 → 3 bar
Sample	Powdered catalyst gently pressed in the cavity
XAS measure	Fluorescence
Multimodal	Raman (UV-vis, IR,...)

La Fontaine *et al.*, *Cat. Today*, 2013, **205**, 148-158

Thermal catalysis
Lytle-type cell

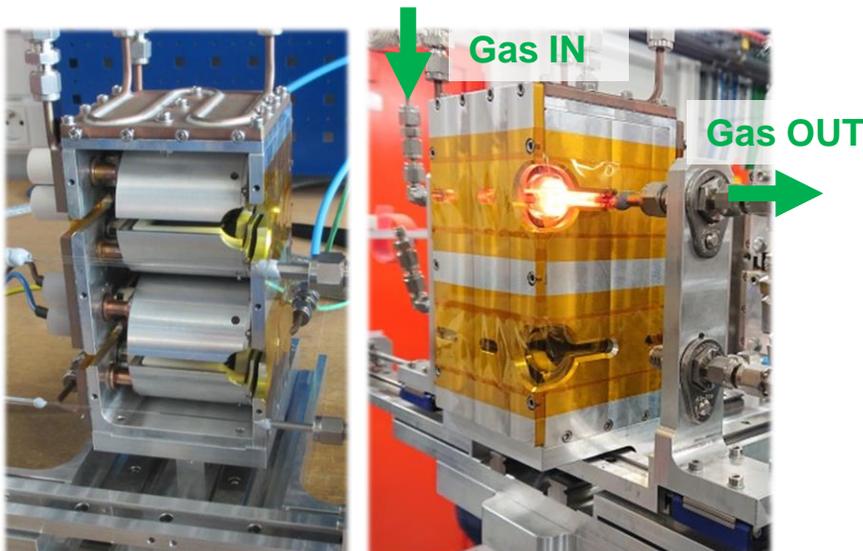


Temperature	Room temp. → 600°C
Pressure	1 → 50 bar
Sample	Powdered catalyst gently pressed in the cavity
XAS measure	Transmission
Multimodal	XRD

La Fontaine et al., *Cat. Today*, 2013, **205**, 148-158

Thermal catalysis Capillary cell

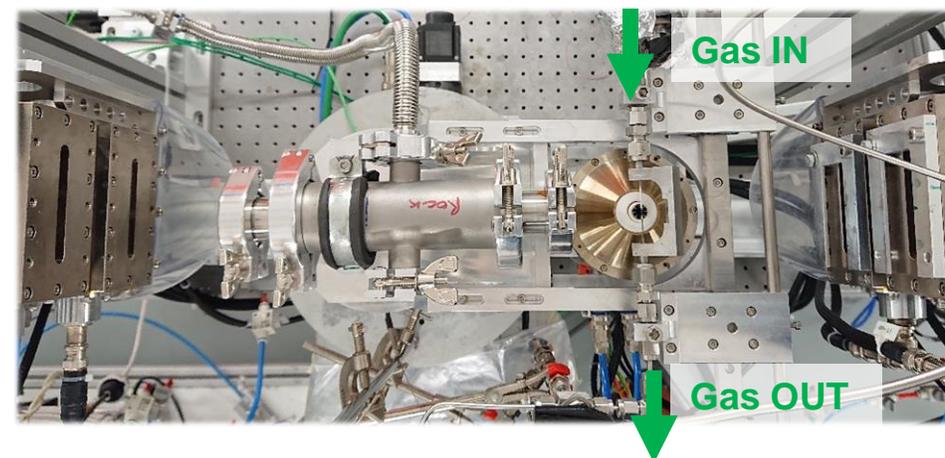
« Four 1000 »



- Temperature control more precise
- Fluorescence and other technique are difficult

Gas blower

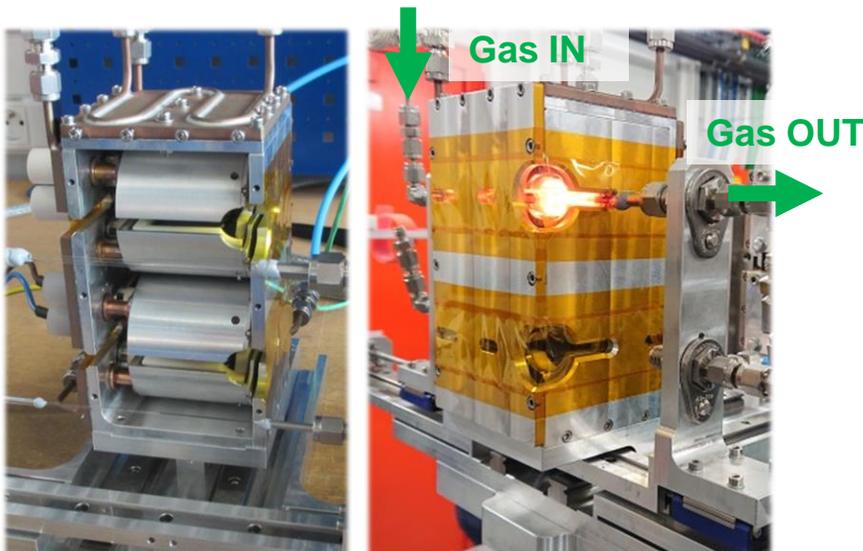
- Fluorescence possible
- Coupling techniques possible
- Temperature control not optimal



Temperature	Room temp. → 1000°C
Pressure	1 → 20 bar
Sample	Powdered catalyst in a cap. between glass-wool
XAS measure	Transmission/fluorescence
Multimodal	XRD, Raman,...

Thermal catalysis Capillary cell

« Four 1000 »



- Temperature control more precise
- Fluorescence and other technique are difficult

Temperature	Room temp. → 1000°C
Pressure	1 → 20 bar
Sample	Powdered catalyst in a cap. between glass-wool
XAS measure	Transmission/fluorescence
Multimodal	XRD, Raman,...

Size and wall thickness of quartz capillaries?

- Wall thickness? → Depends mainly on the energy
- Size of cap. diameter? → Depends on the energy and the catalyst composition

<https://11bm.xray.aps.anl.gov/absorb/absorb.php>

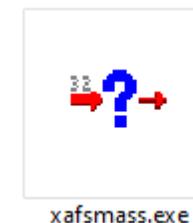
Compute X-ray Absorption. This tool is currently under development.

Select X-ray Wavelength or Energy: [\(click for details\)](#)
 Energy (keV)

Chemical Formula: [\(click for details\)](#)
 enter using element chemical symbol and formula unit occupancy, e.g. YBa2Cu3O6.5 (f.u.)

Sample Radius: [\(click for details\)](#)
 capillary radius in mm

Sample Density or Packing Fraction [\(click for details\)](#)
 enter measured sample density or estimated packing fraction (often ~0.6)
 Packed Fraction (0.0 - 1.0)



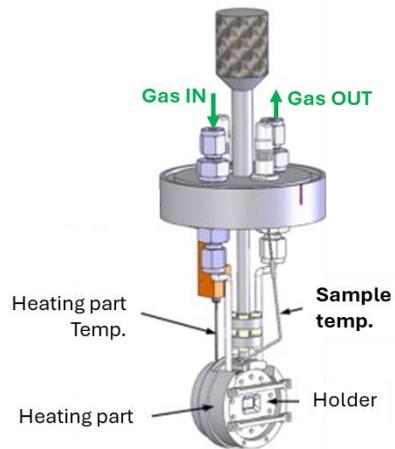
<https://xafsmass.readthedocs.io/#>

Sometimes, dilution is needed...

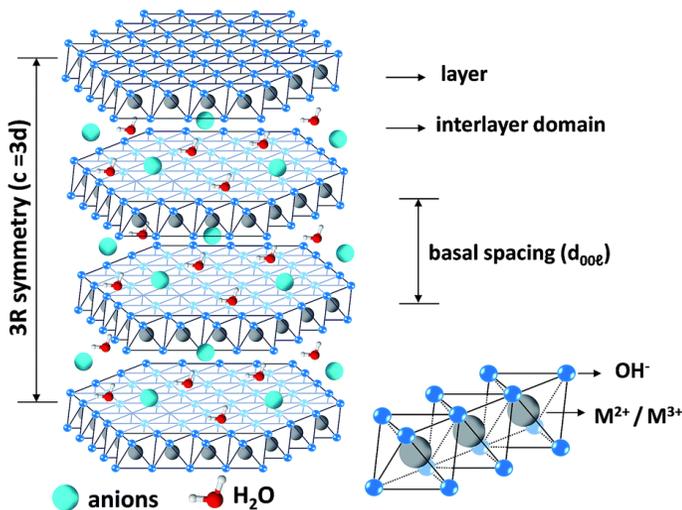
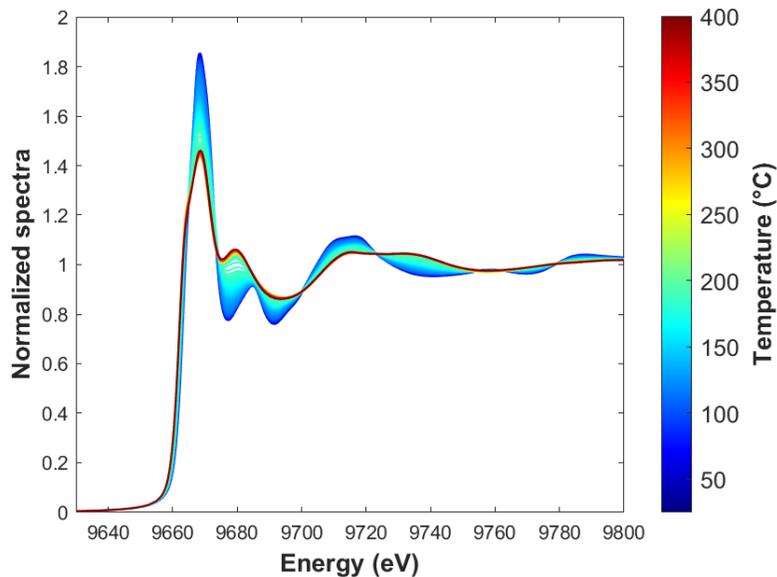
- Dilutant not too much absorbent
- Dilutant inert regarding the reaction of interest
- Take care of dilutant used if coupling techniques
- Dilution must be considered regarding the activity

Reactions

Ethanol Steam Reforming
 Dry Methane reforming
 Fischer Tropsch
 HDS & HDO
 CO₂ – CO hydrogenation
 Chemical Looping Combustion
Memory effect of LDH

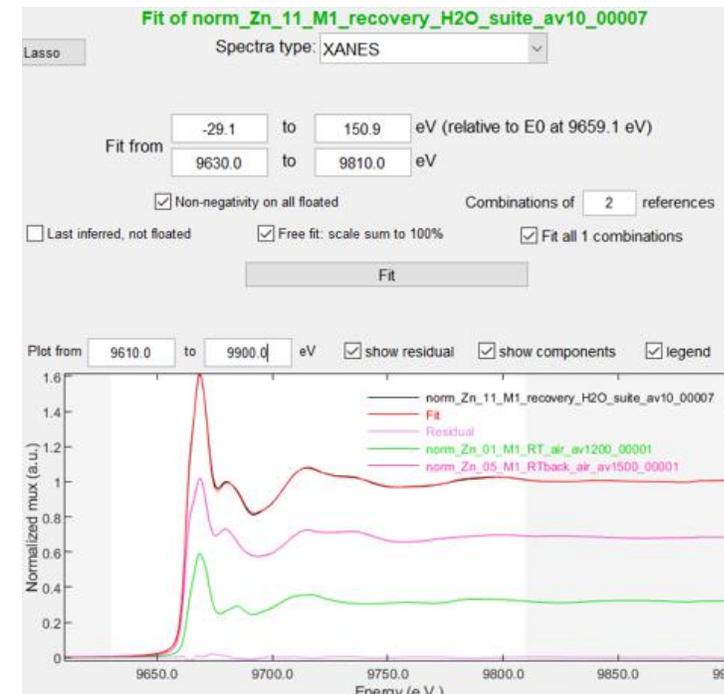


1/ Heating to 400°C under air flow



Santos *et al.*, *J. Matter Chem. A.* 2017, **5**, 9998

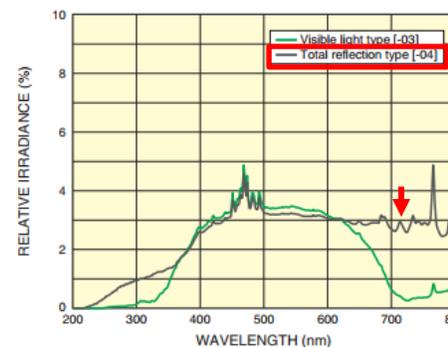
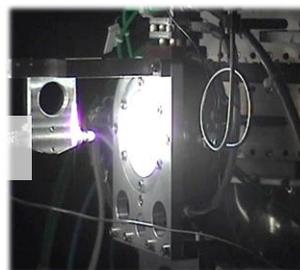
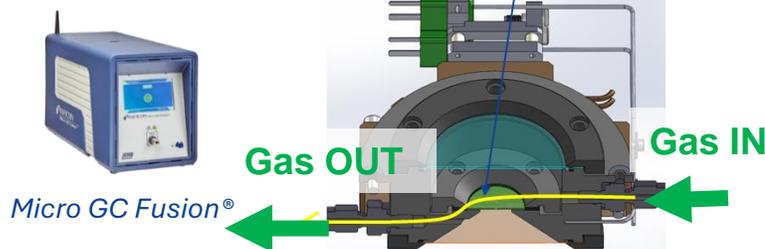
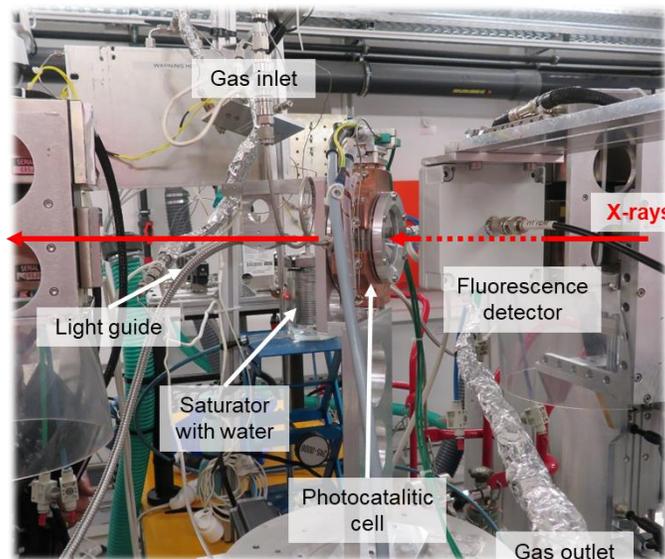
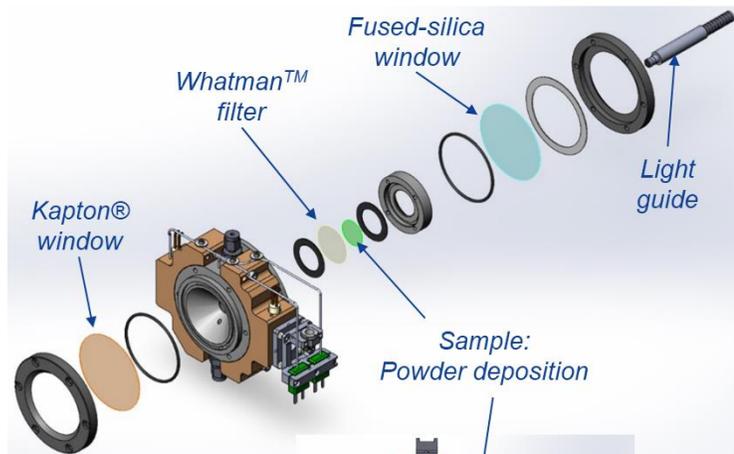
2/ Adding NaCl solution at RT



Fit 1: R factor= 2.819895e-05, Chi-square= 0.01370394, Reduced chi-square= 2.399989e-05
 Sum of all fractions= 100.0
 norm_Zn_01_M1_RT_air_av1200_00001= 32.0
 norm_Zn_05_M1_RTback_air_av1500_00001= 68.0

- LDH is decomposed into nano-ZnO
- Recovery of ~35 % of the starting LDH after adding NaCl solution

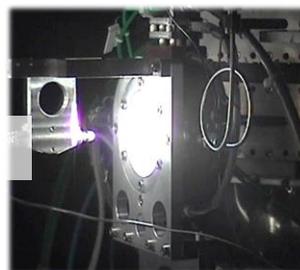
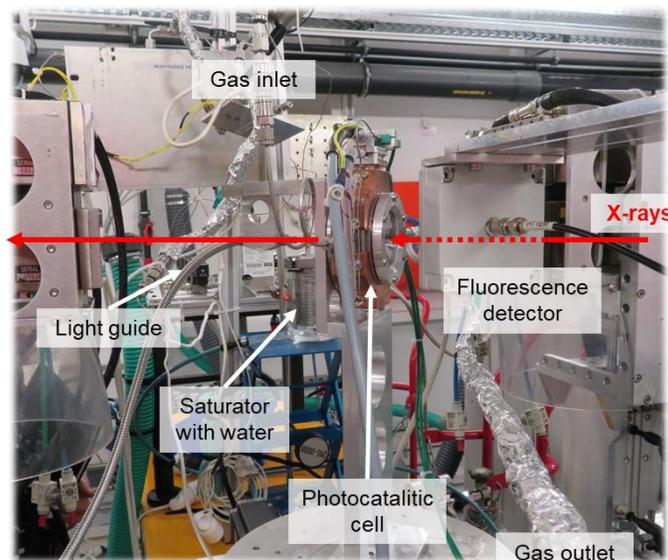
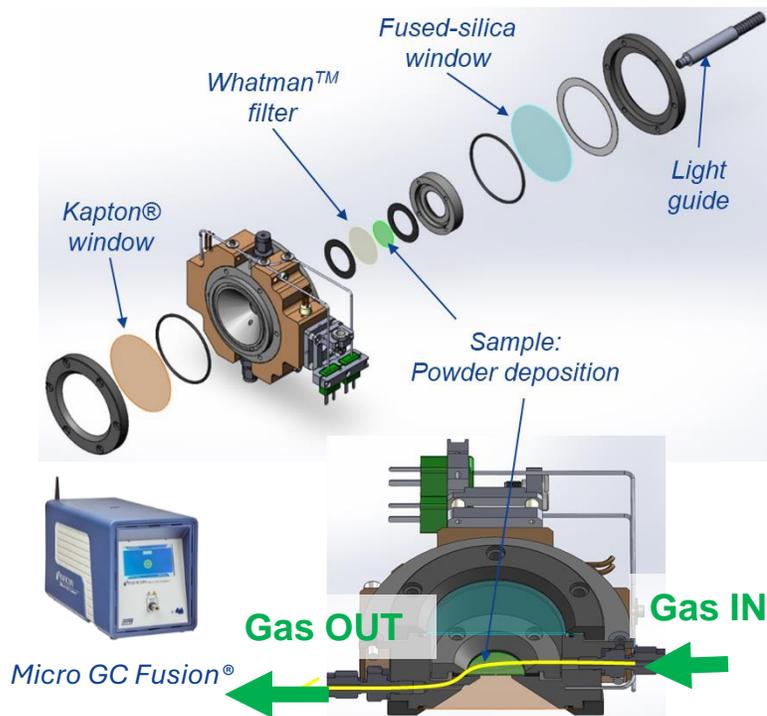
Photo-catalysis Gas-phase reaction



It is crucial to verify that the activity of the material in the *operando* cell is the same than at laboratory

Temperature	Room temp. → 80°C
Pressure	1 bar
Sample	Powdered catalyst gently pressed/pellet
XAS measure	Transmission/fluorescence
Multimodal	Not performed yet

Photo-catalysis Gas-phase reaction

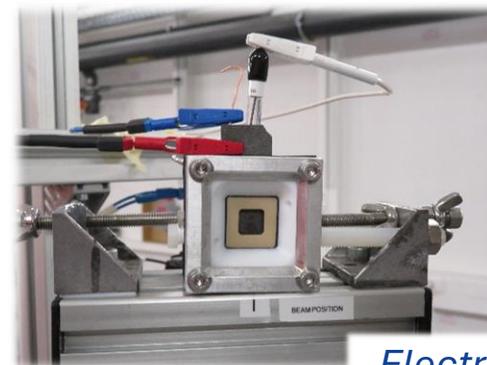
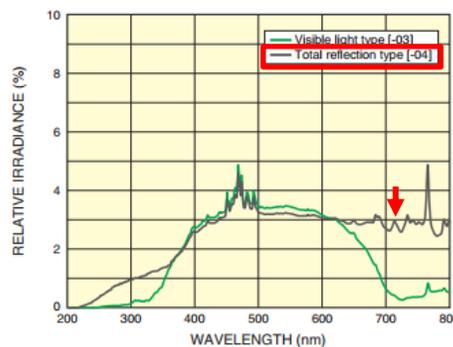
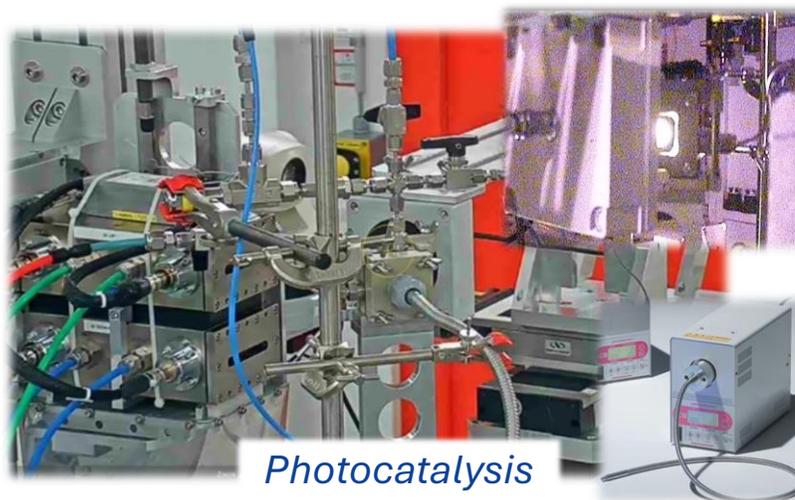
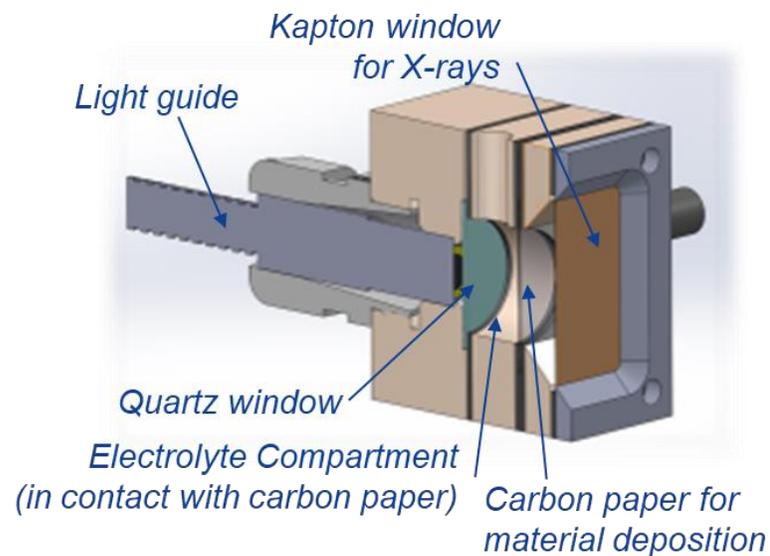


Reactions

CO₂ photoconversion
Ethanol photoconversion
Formic Acid for H₂ production

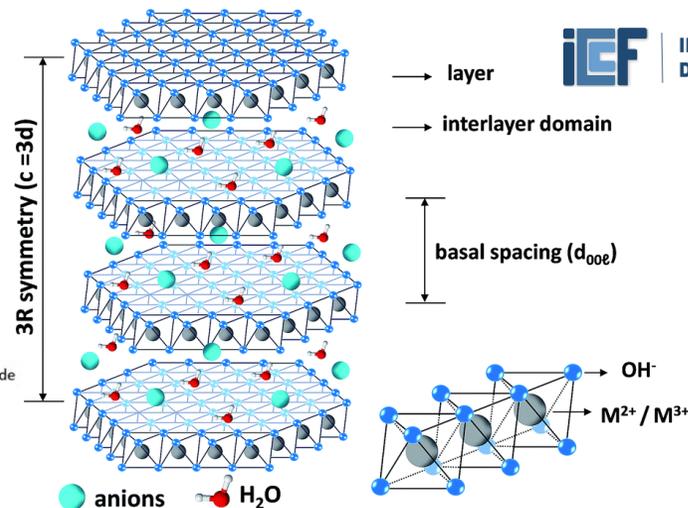
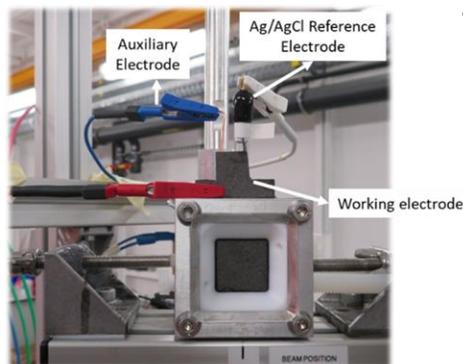
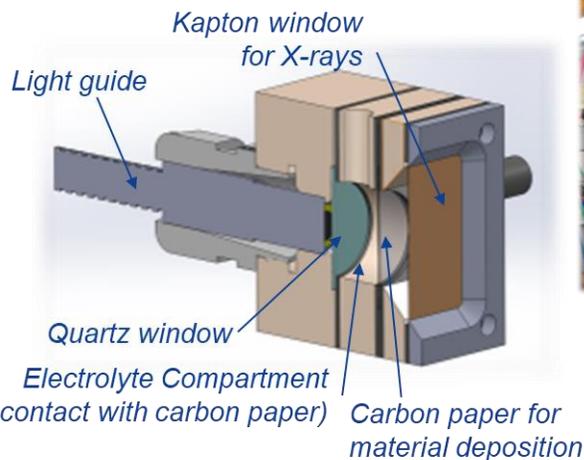
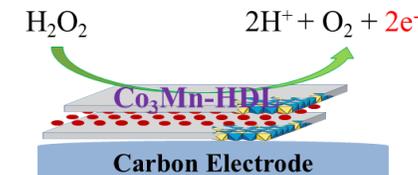
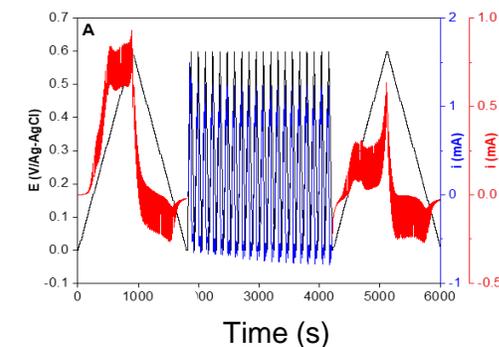
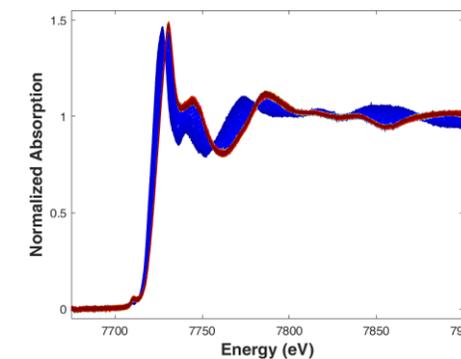
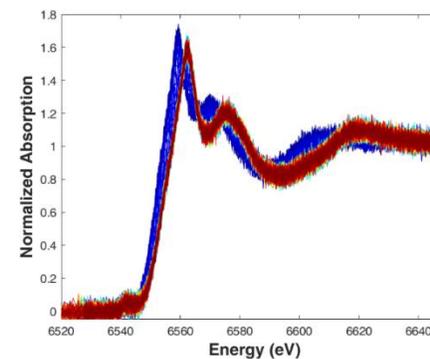
Study not published

Temperature	Room temp. → 80°C
Pressure	1 bar
Sample	Powdered catalyst gently pressed/pellet
XAS measure	Transmission/fluorescence
Multimodal	Not performed yet

(Photo)-(electro)-catalysis
Gas-phase reaction

Temperature	Room temperature
Pressure	1 bar
Sample	Sample deposited on Carbon Paper
XAS measure	Fluorescence
Multimodal	Not performed yet

The more versatile, the better

(Photo)-(electro)-catalysis
Gas-phase reactionSantos et al., *J. Matter Chem. A.* 2017, 5, 9998Farhat et al., *J. Phys. Chem.*, 2020, 124, 15585-15599**Reactions**Photocatalytic water splitting
LDH electroactivity**Transformation during electro-chemical CV activation**Farhat et al., *J. Phys. Chem. C.* 2024, 128, 21023

Temperature	Room temperature
Pressure	1 bar
Sample	Sample deposited on Carbon Paper
XAS measure	Fluorescence
Multimodal	Not performed yet

I. Examples of *in situ/operando* setups

Catalysis (thermal catalysis, (photo)-(electro)-catalysis)

Energy storage

Liquid cells

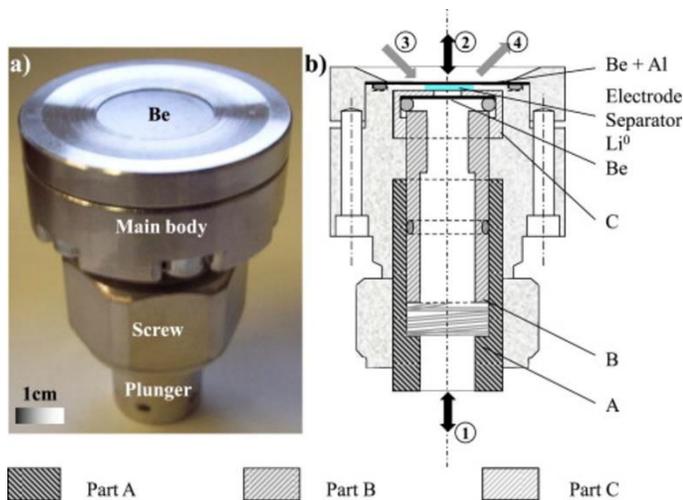
II. Multimodal characterisations and data analysis methodology

III. Scientific case: LDH used for Ethanol Stream Reforming

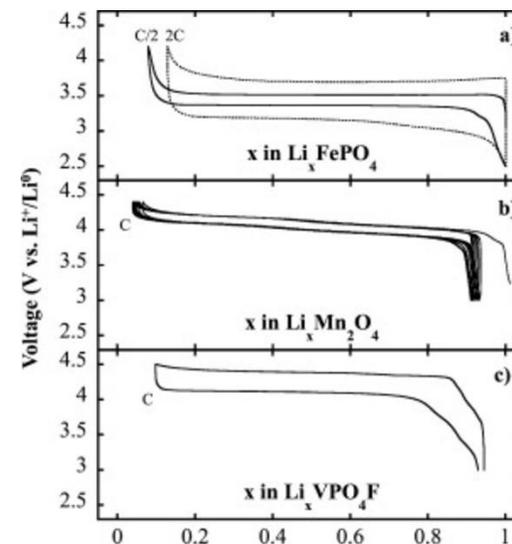
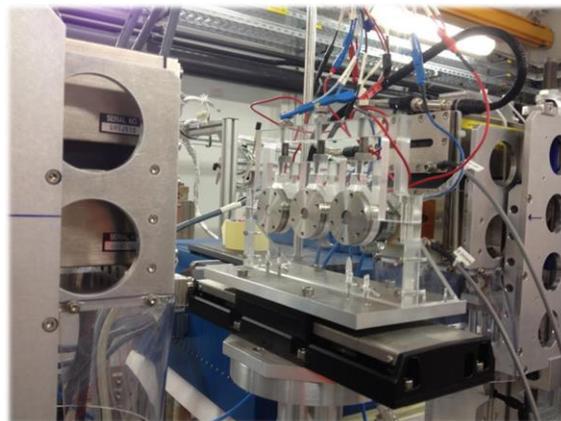
IV. Conclusions

An example of cell used for energy storage

Specific cell for cycling battery materials



Setup on a beamline

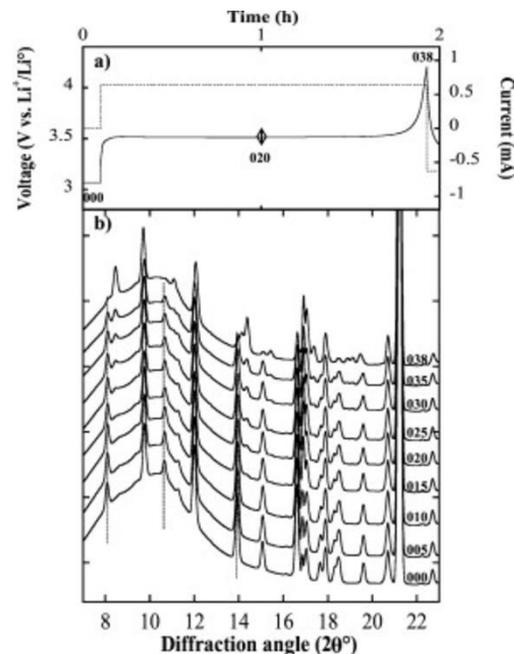


Electrochemistry validated for *several systems*

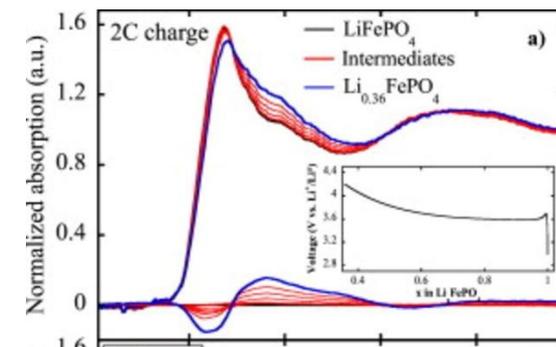
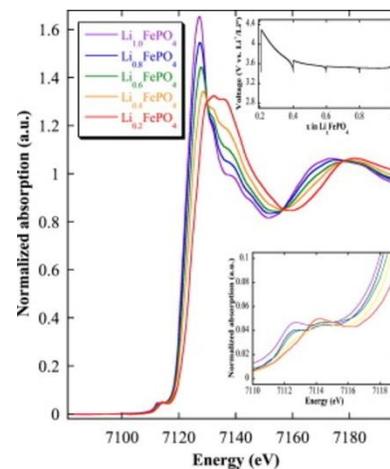
Potentiostat to apply potential and measure current



XRD characterisation



XAS characterisation



I. Examples of *in situ/operando* setups

Catalysis (thermal catalysis, (photo)-(electro)-catalysis)

Energy storage

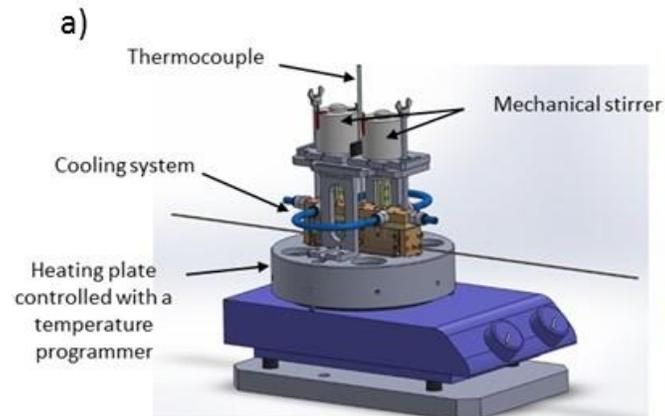
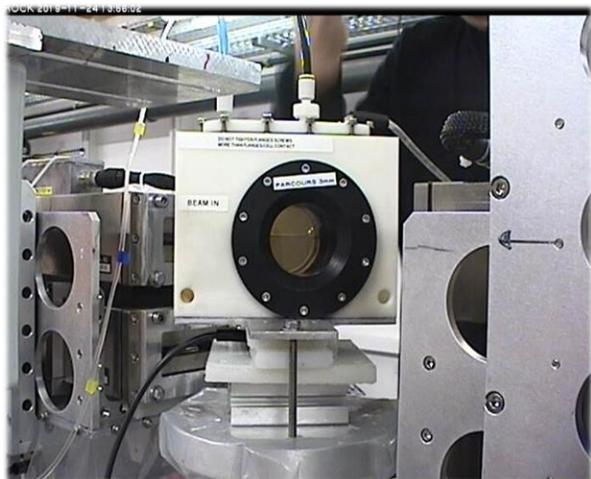
Liquid cells

II. Multimodal characterisations and data analysis methodology

III. Scientific case: LDH used for Ethanol Stream Reforming

IV. Conclusions

Liquid cells



Azeredo et al., PCCP, 2023, 25, 22523



Temperature	Room temperature → 180°C
Pressure	1 bar
Sample	Liquids (concentration to be optimized)
XAS measure	Fluorescence/Transmission
Multimodal	Raman, UV-visible,...

- Several optical paths are possible
- Several types of windows are possible

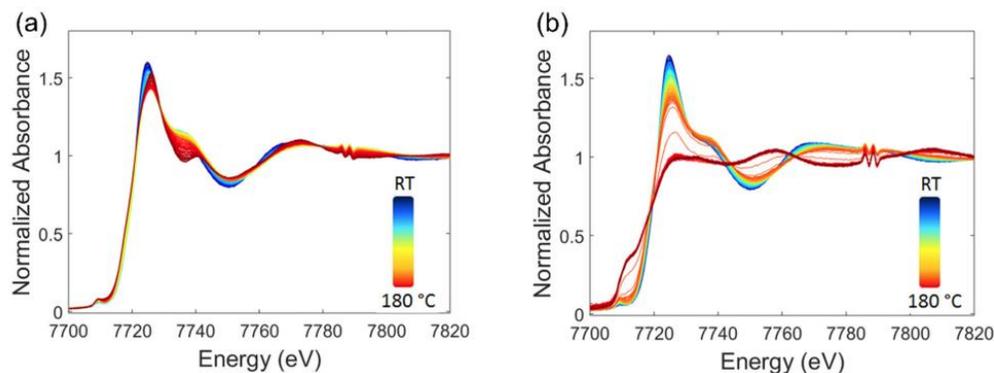
Liquid cells

Reactions

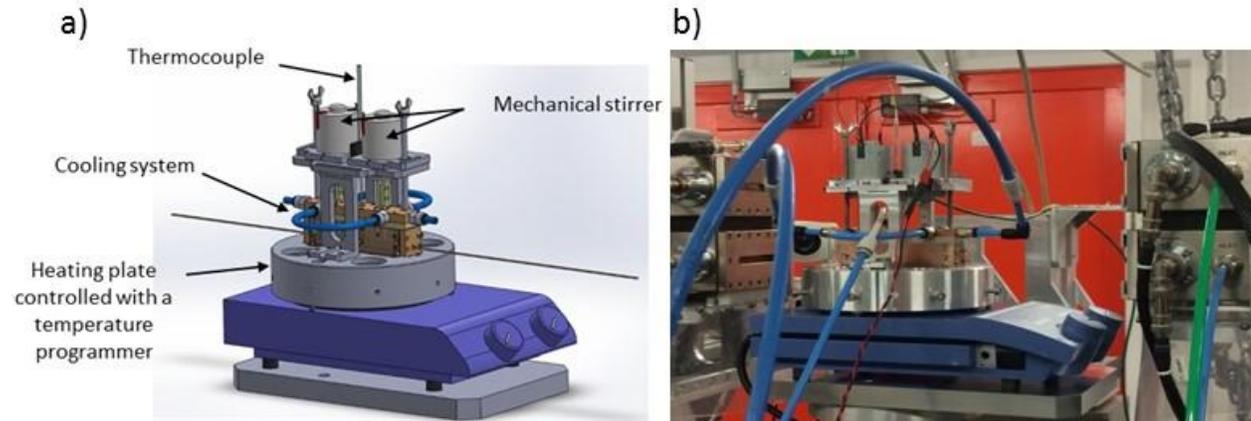
Nucleation growth of nanoparticles
LDH formation
Determination of species in solution



Formation of Co-Ru alloys



Temperature	Room temperature → 180°C
Pressure	1 bar
Sample	Liquids (concentration to be optimized)
XAS measure	Fluorescence/Transmission
Multimodal	Raman, UV-visible,...



Azeredo et al., PCCP, 2023, 25, 22523



- Several optical paths are possible
- Several types of windows are possible

- ✓ **What is the application?**
 - Example: photocatalysis? → Example: quartz window
- ✓ **Are there any other coupling methods?**
 - Example: Raman? → Example: mica window
- ✓ **What is the transmission of X-rays through the window material?**
 - Use website/software for calculation, for instance https://henke.lbl.gov/optical_constants/



- X-Ray Database
- Nanomagnetism
- X-Ray Microscopy
- EUV Lithography
- EUV Mask Imaging
- Reflectometry
- Zoneplate Lenses
- Coherent Optics
- Nanofabrication
- Optical Coatings
- Engineering

Tell us what else you wish this tool could do.

We want to make this tool even more capable and useful to you. Let us know how it can be improved.

X-Ray Interactions V

Introduction

Access the [atomic scattering factor](#) files.

Look up [x-ray properties of the elements](#).

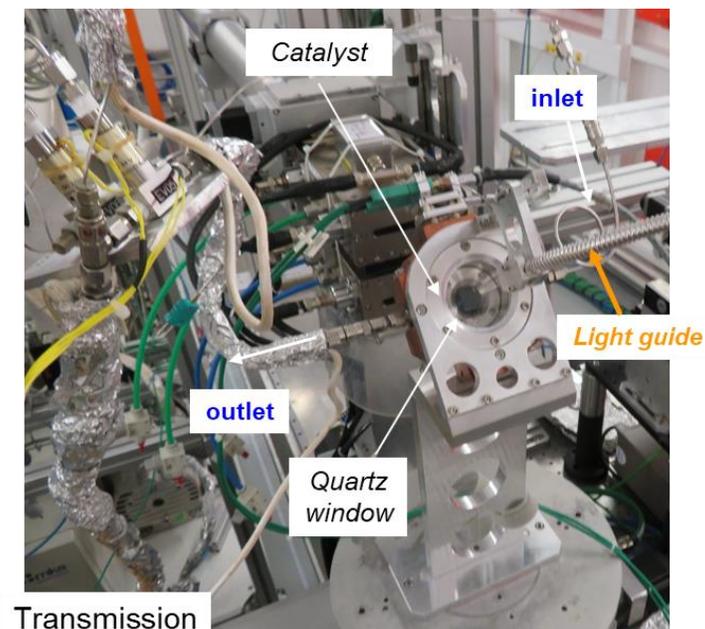
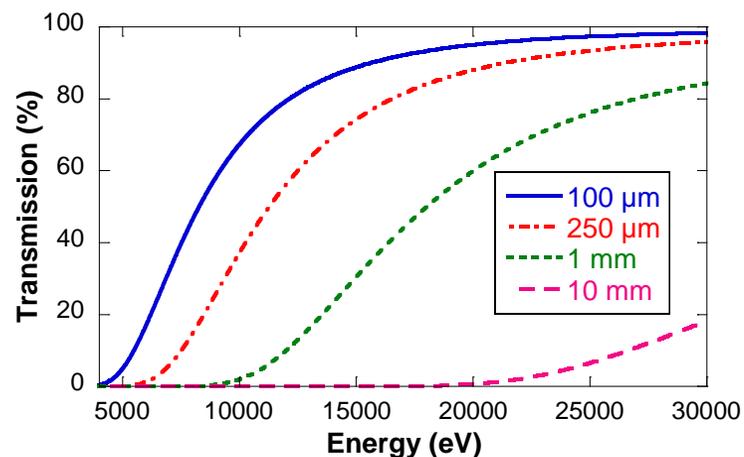
The [index of refraction](#) for a compound material.

The [x-ray attenuation length](#) of a solid.

X-ray transmission

- Of a [solid](#).
- Of a [gas](#).

Transmission of SiO₂



Transmission

- Mo K-edge (20 keV) → 1 mm possible
- Cu K-edge (~9 keV) → 100 μm better

• Choose from a list of common materials:

• Chemical Formula:

• Density: gm/cm³ (enter negative number to use tabulated values.)

• Thickness: microns

• Photon Energy (eV) Range from to in steps (< 1000).

(NOTE: Photon Energy must be in the range 10 eV < E < 30,000 eV and Wavelengths in the range 40 Å > λ > 0.4 Å)

request a press this button:

reset to default values, press this button:

- ✓ **Is there any incompatibility between the material in which the cell is made and the material of interest?**
 - Especially for fluorescence measurements

- ✓ **Is the window for X-rays suitable for the working energy?**
 - Material
 - Thickness

- ✓ **Does the cell allow to perform the reaction in real conditions?**
 - Temperature
 - Pressure
 - Measurement of the activity of the material

- ✓ **Is the activity of the material in the cell similar to the one observed at the lab?**
 - Need to perform test before the *operando* measurements

- ✓ **Is the cell compatible with other characterization techniques?**
 - Raman, XRD, UV-vis,...

I. Examples of *in situ/operando* cells

II. Multimodal characterisations and data analysis methodology

UV-visible coupling

Raman coupling

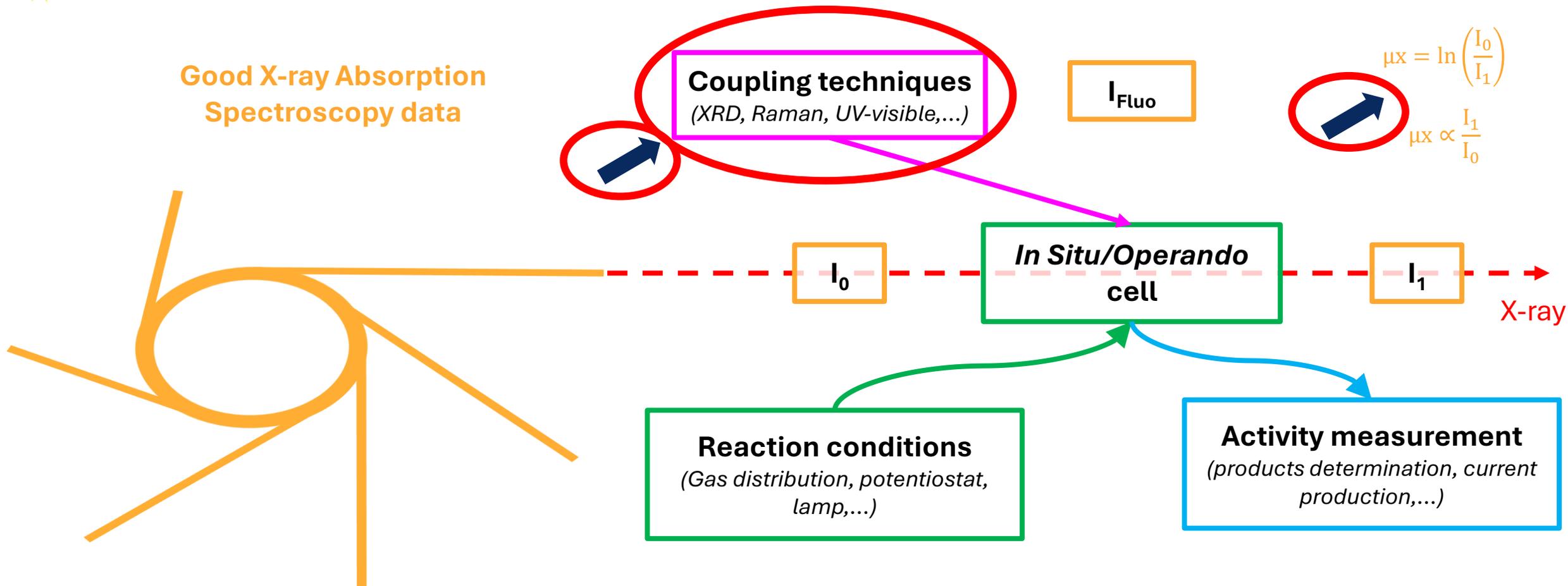
X-Ray Diffraction coupling

III. Scientific case: LDH used for Ethanol Stream Reforming

IV. Conclusions

What is needed for *operando* characterization using XAS?

Good X-ray Absorption Spectroscopy data



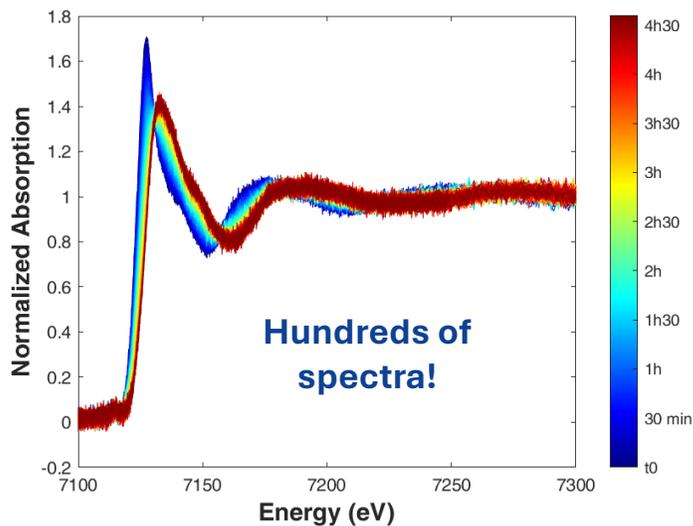
Cells able to reproduce real reaction conditions

Qualitative and Quantitative access to the material activity

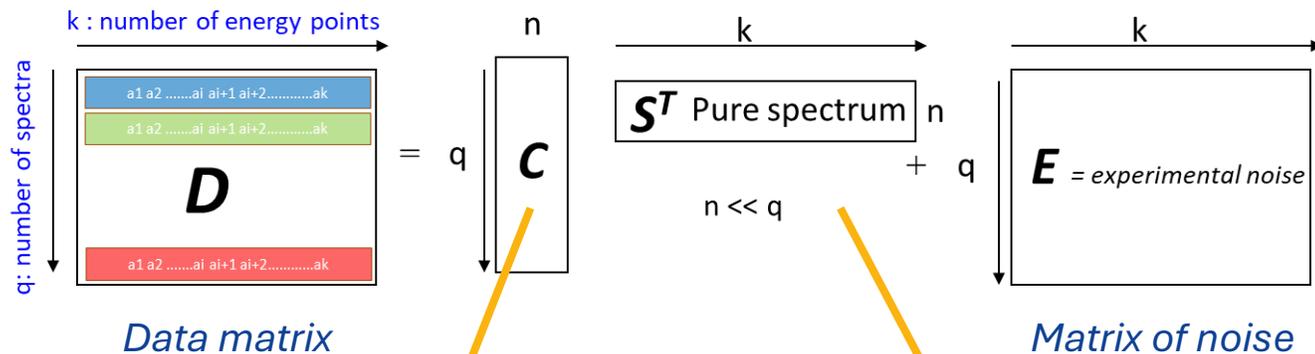
Complementary Methods for simultaneous Characterization

Large dataset analysis methodology

Multivariate Curve Resolution Alternating Least Square fitting

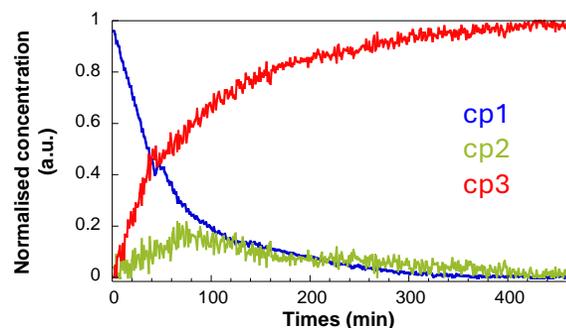


Radiation decomposition of Gold nanoparticles
Study not published
Example modified

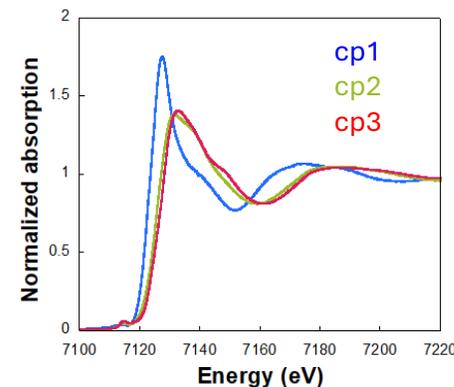


XAS: Beer-Lambert Law $\rightarrow Abs = \sum c_i Abs_i$
 \Rightarrow Bilinear decomposition

- Several examples (XAS applications):
 Catalysis Today (2014) 229, 114-122
 Comptes Rendus Chimie (2016) 19, 1337-1351
 J. Phys. Chem. C (2017) 121, 18544-18556
 Catalysis Today (2019) 336, 63-73
 PCCP (2020) 22, 18835-18848

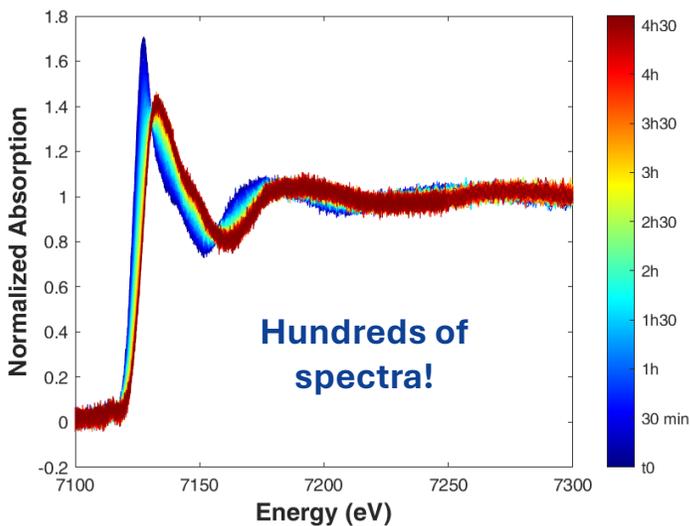


Concentration profile



Pure spectra

Multivariate Curve Resolution Alternating Least Square fitting



Step 1: determination of the number of Components required to explain the variance of the D matrix

Principal Component Analysis

Radiation induced degradation of Gold nanoparticles
Study not published
Example modified



$$D = CS^T + E$$

Step 2: Initial estimation of the pure spectra (S^T)



Step 4: ALS minimization

$$\min \|D - CS^T\|^2$$



4a. Determination of concentration matrix
 4b. The new matrix is used as a new guess

Step 3: Define some constraints to help the minimization

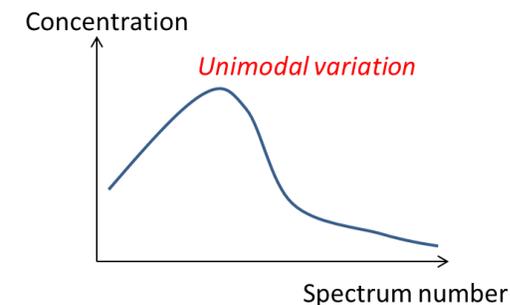
✓ Non-negativity of C and S

✓ Closure relation on C

The sum of the concentration of each component for each spectra of the matrix must be 1

$$\sum_{i=1}^n c_i = 1$$

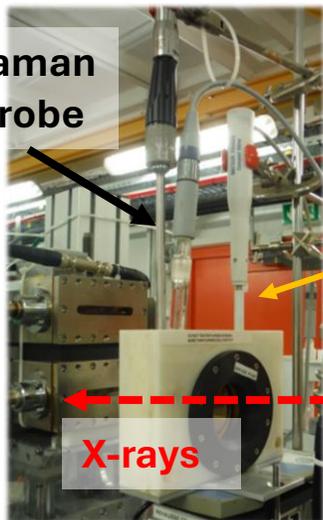
✓ Unimodality of concentration profile (optional)



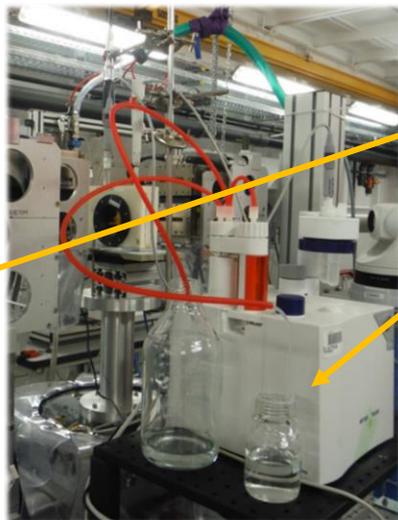
Iteration is stopped when convergence is achieved

In Situ

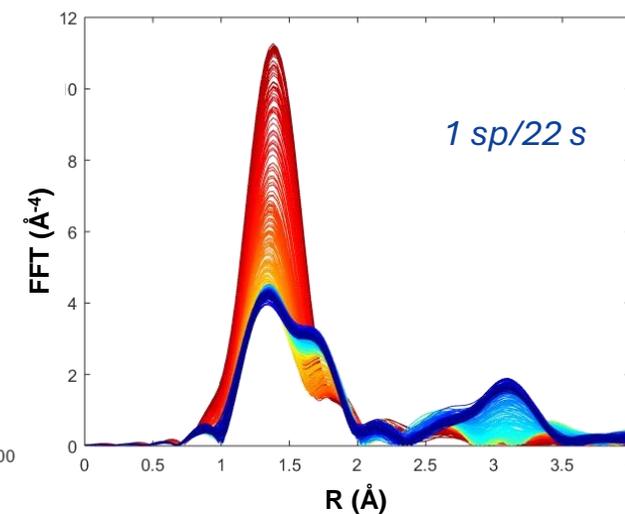
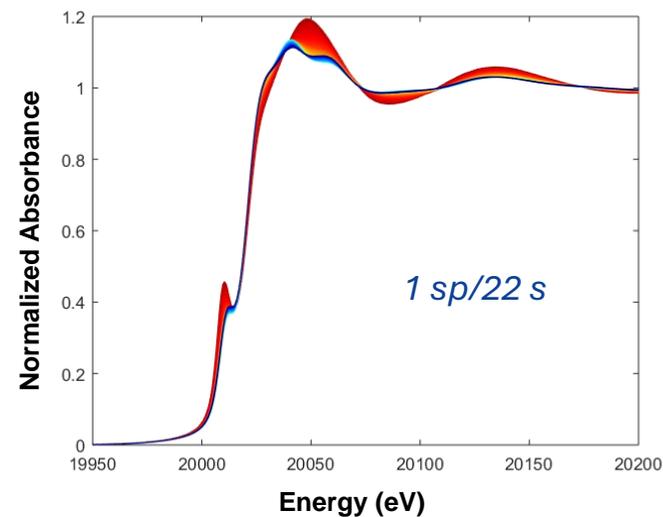
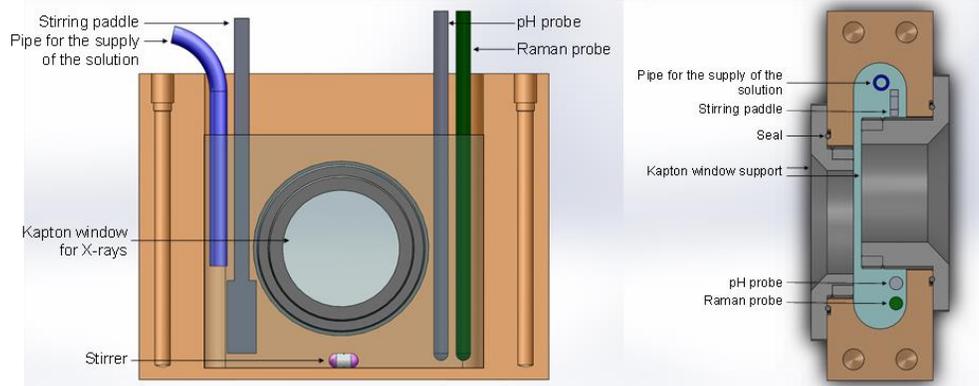
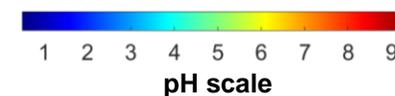
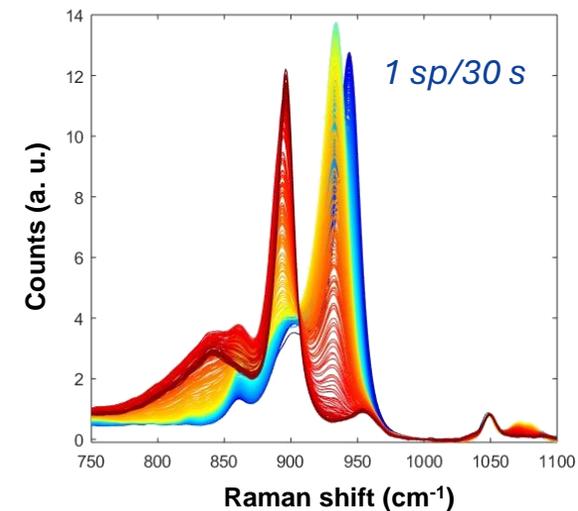
Determination of Mo species in solution

Raman
probe

X-rays

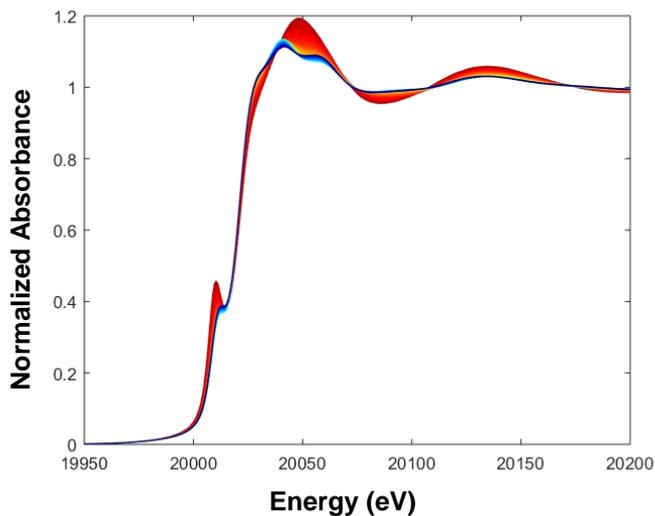


- HMA + Citric Acid (CA)
- $[Mo]=[CA] = 2 \text{ mol L}^{-1}$
- $\text{pH}_{\text{init}} = 0.58$
- NaOH addition:
 0.25 mL min^{-1} (pH 0.58 \rightarrow 7.5)
 0.5 mL min^{-1} (7.5 \rightarrow 9.5)

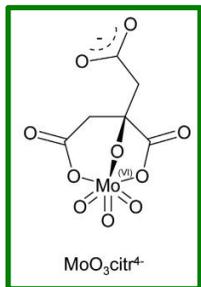
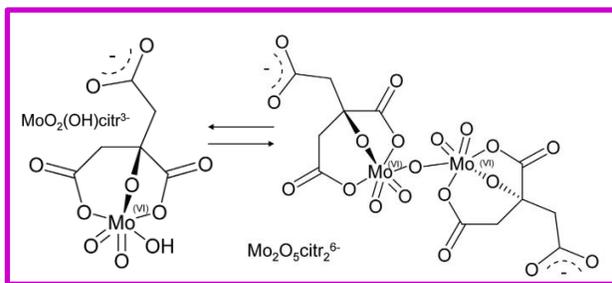
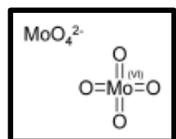
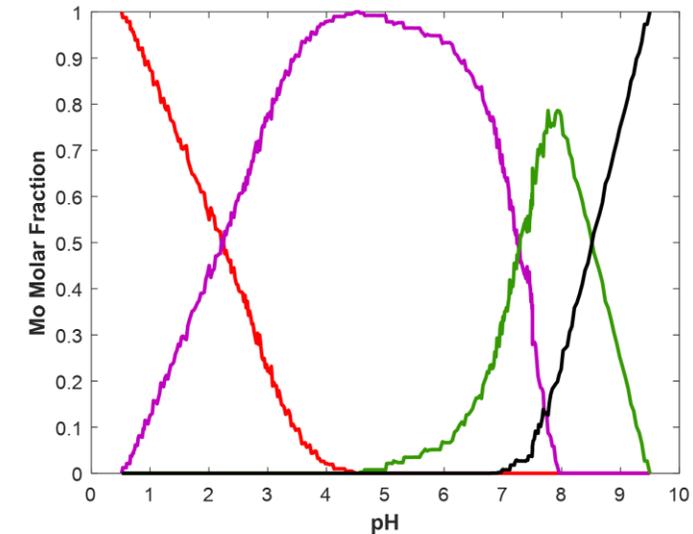
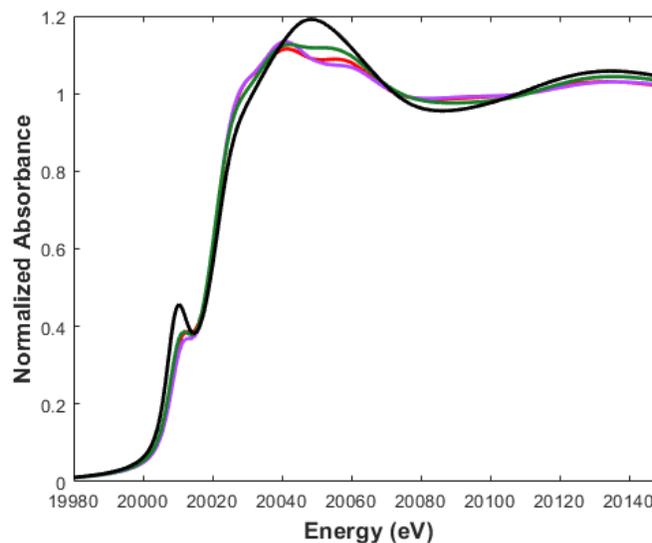


Determination of Mo species in solution

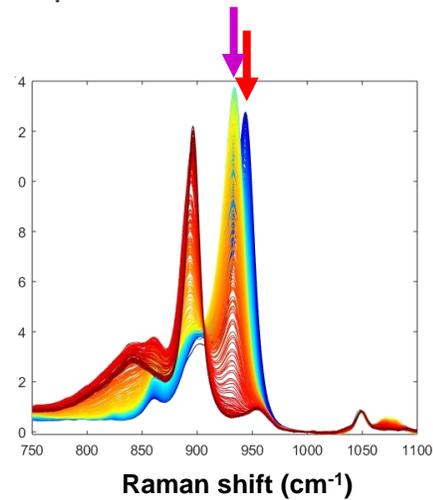
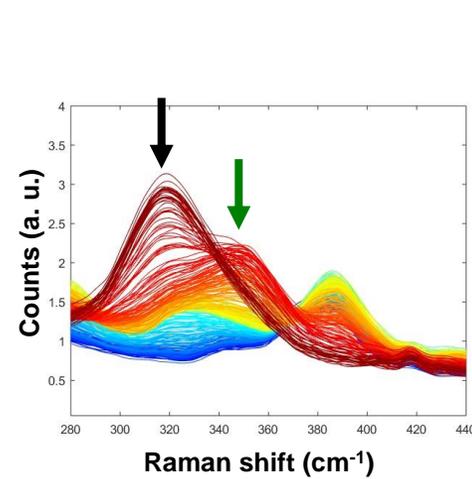
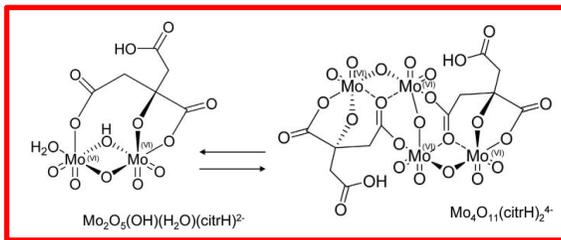
In Situ



Chemometrics



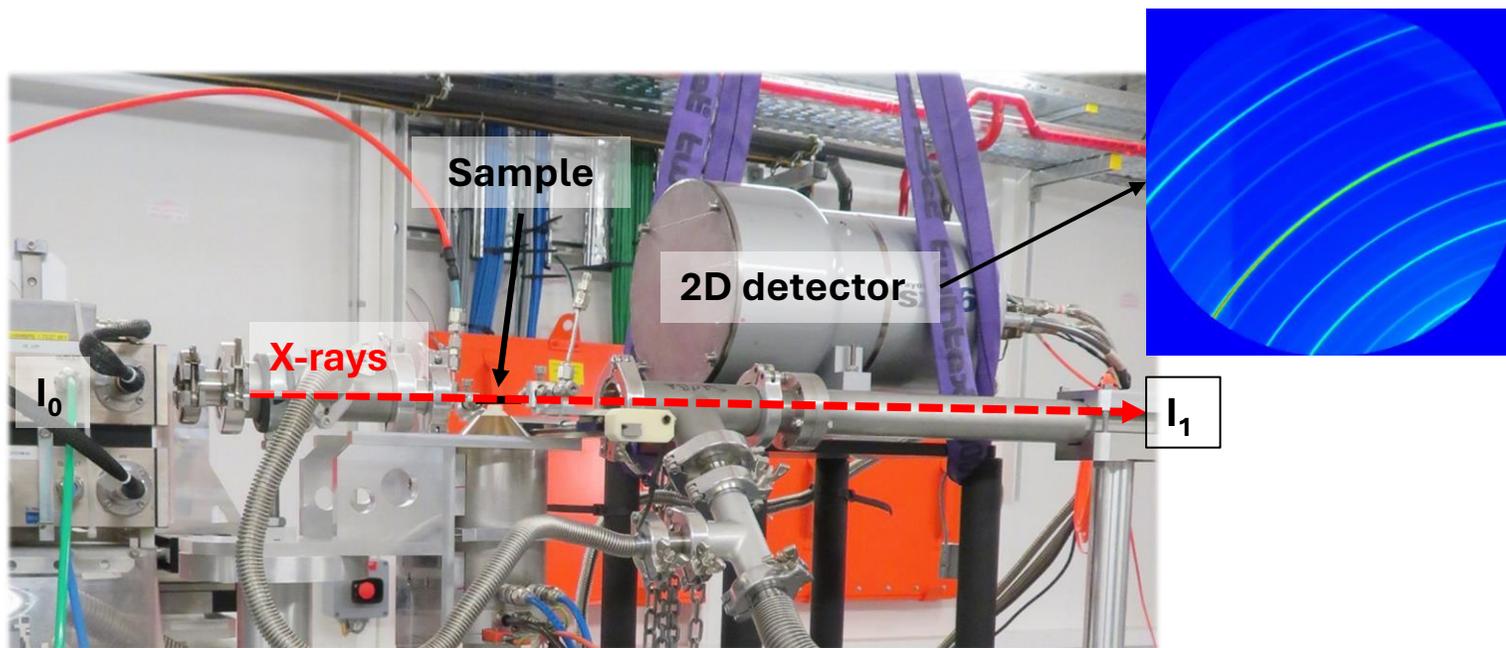
PhD thesis of J.A. Bergwerff (2007)
and P. Sénécal (2013)



Lesage et al., ChemCatChem., 2024, 17

In Situ

High entropy oxides formation upon heating

**Study not published**

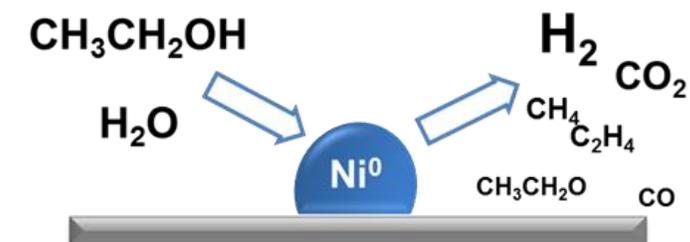
- Precursors:
 - Mechanical mixing of several oxides
- Heating to 900°C at P_{atm} using commercial Gas Blower
- *In situ* monitoring by alternating:
 - X-ray Absorption Spectroscopy
 - X-Ray Diffraction

I. Examples of *in situ/operando* cells

II. Multimodal characterisations and data analysis methodology

III. Scientific case: LDH used for Ethanol Stream Reforming

IV. Conclusions

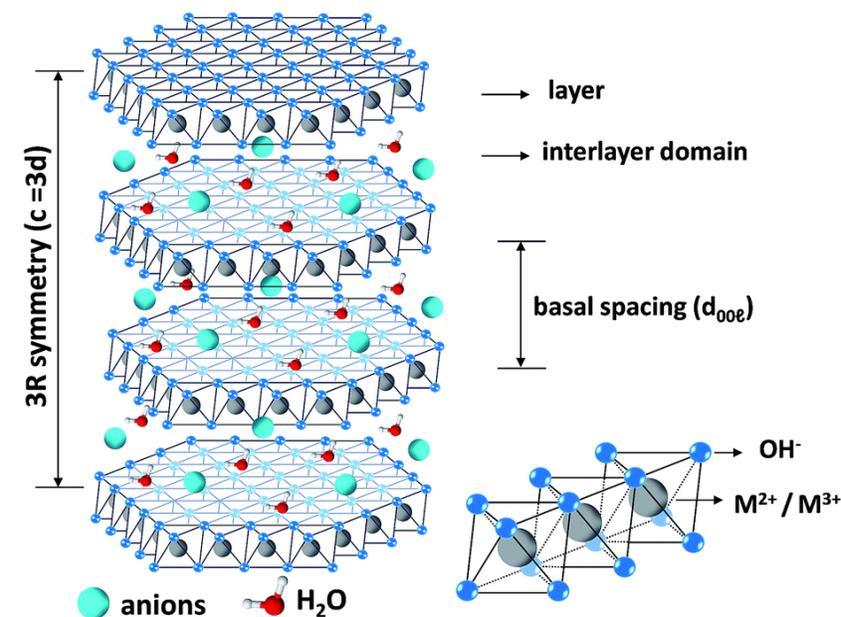


Noble metals
Pd, Pt, Ru, Rh,...

Non noble metals
Co, Ni

Ni: cleavage of C-C and C-H
Cu: ↓ the reduction (activation)
temperature

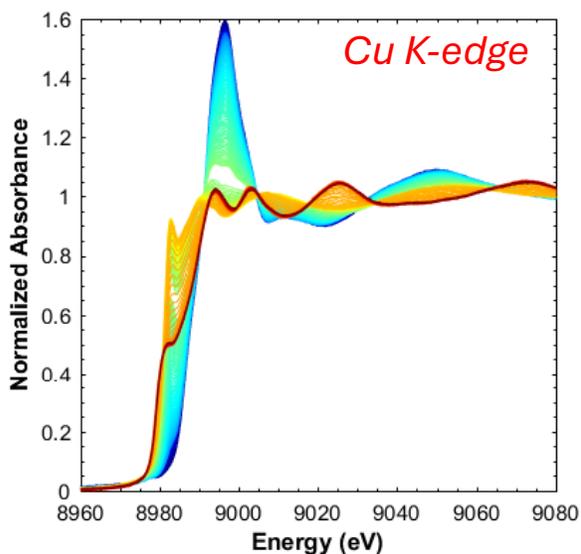
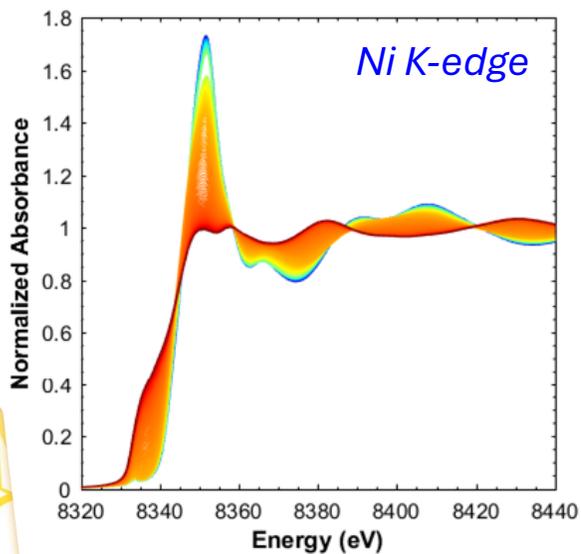
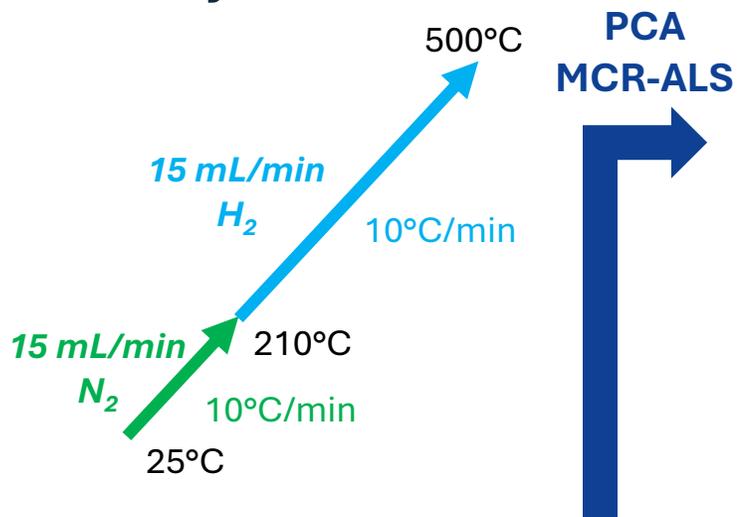
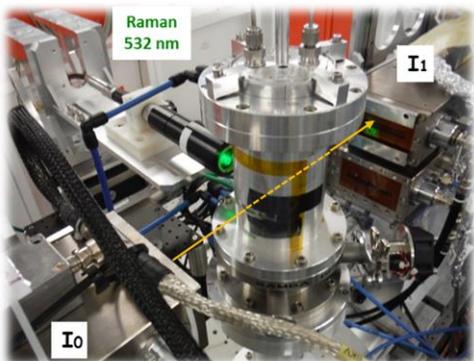
Ni-Cu bimetallic catalysts
LDH structure



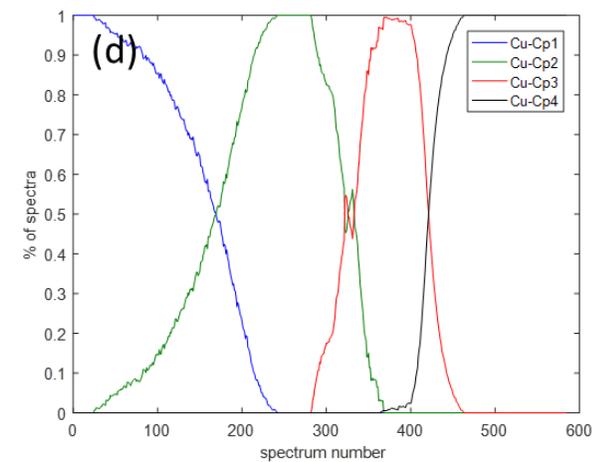
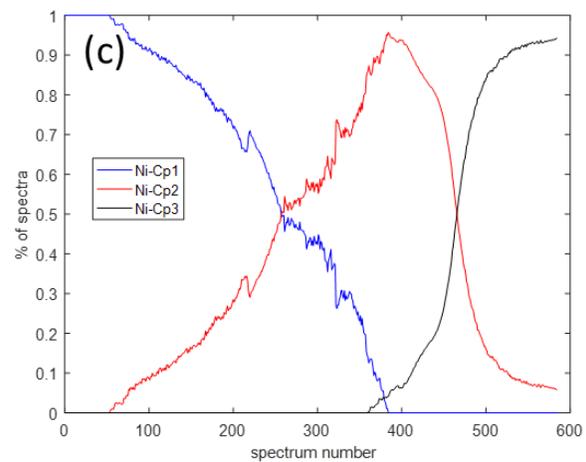
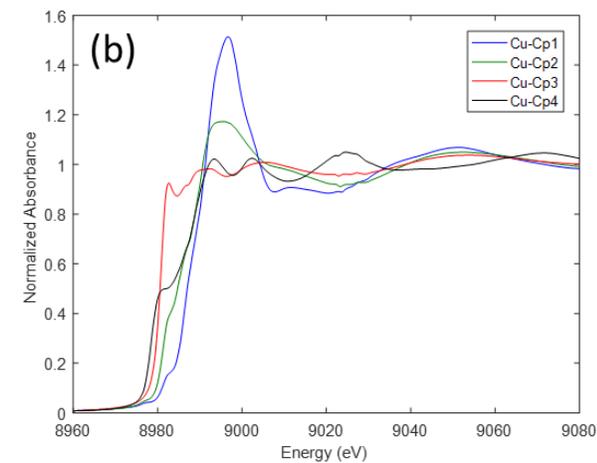
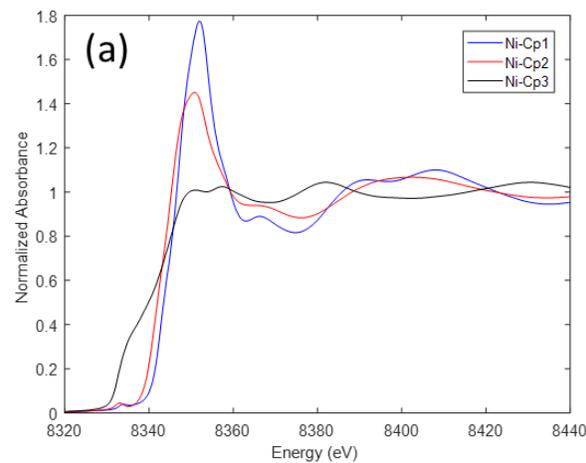
Santos et al., *J. Matter Chem. A.* 2017, 5, 9998

In Situ

Activation of the catalyst



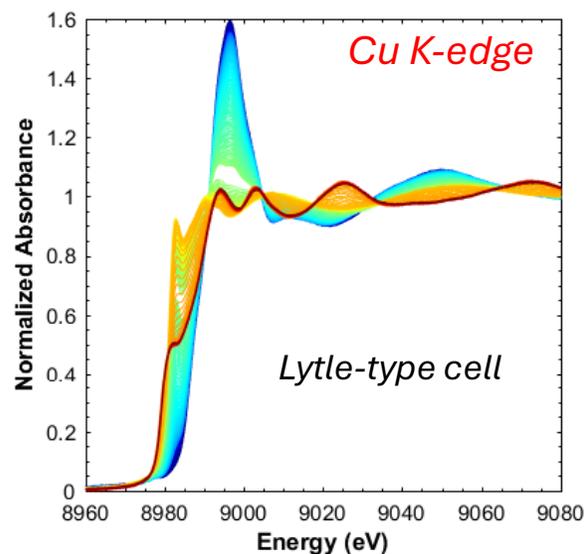
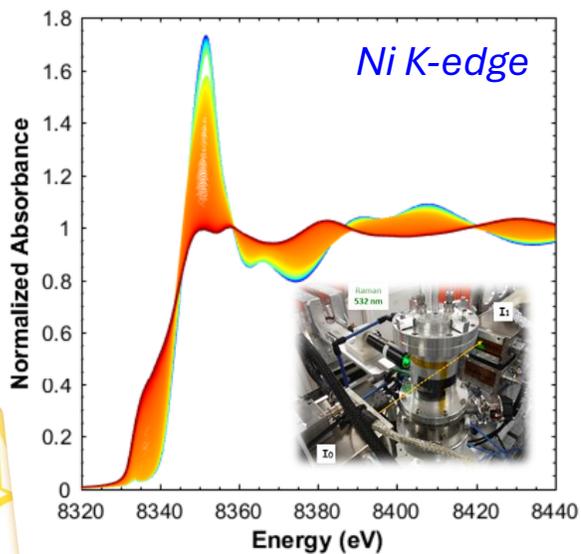
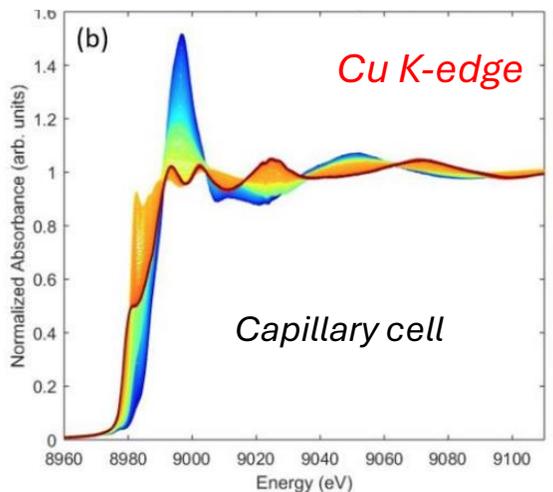
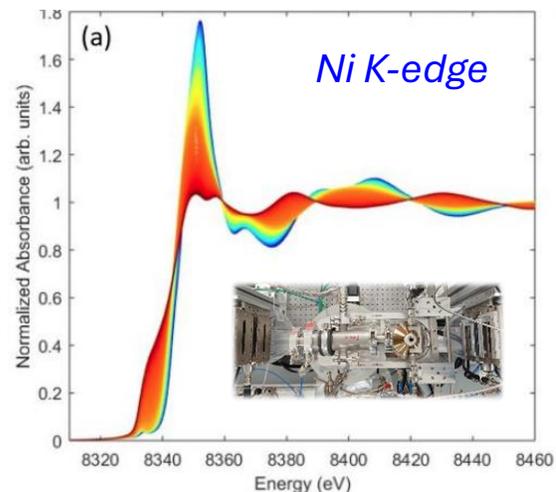
RT 500°C



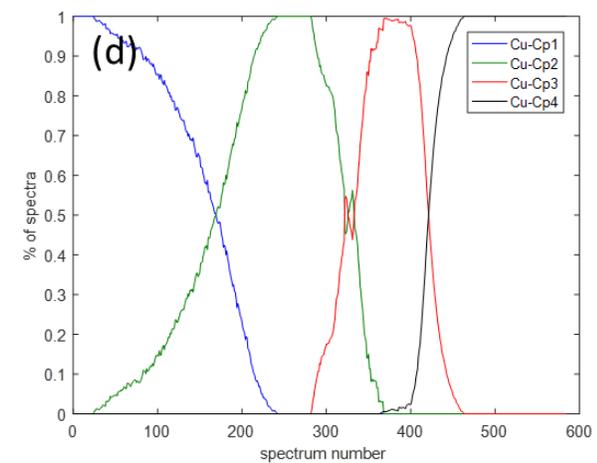
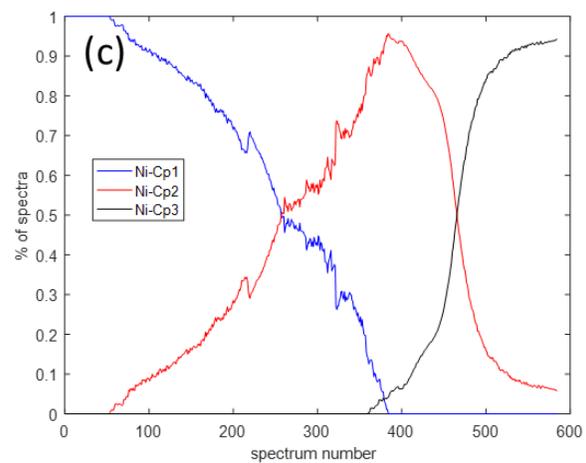
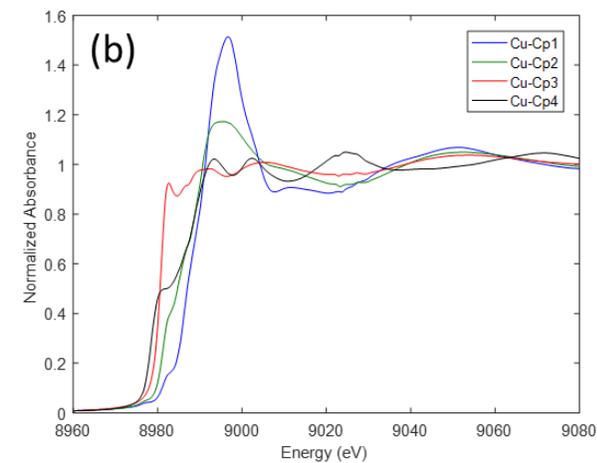
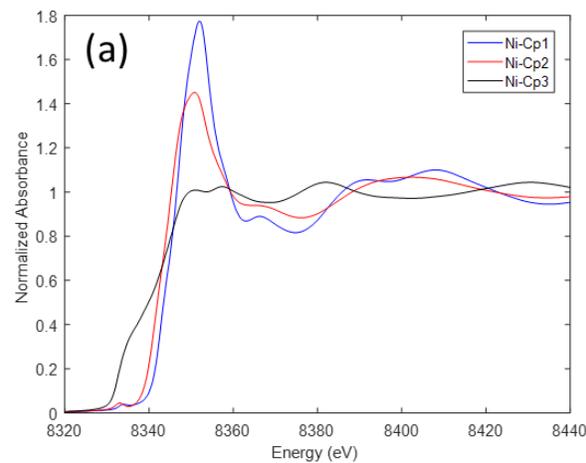
3 components for Ni / 4 components for Cu
Cu reduced quicker than Ni

In Situ

Activation of the catalyst



RT 500°C

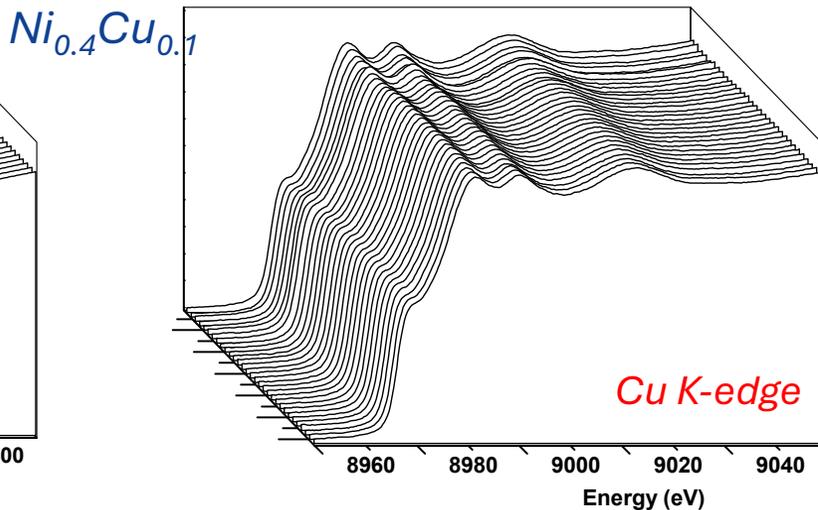
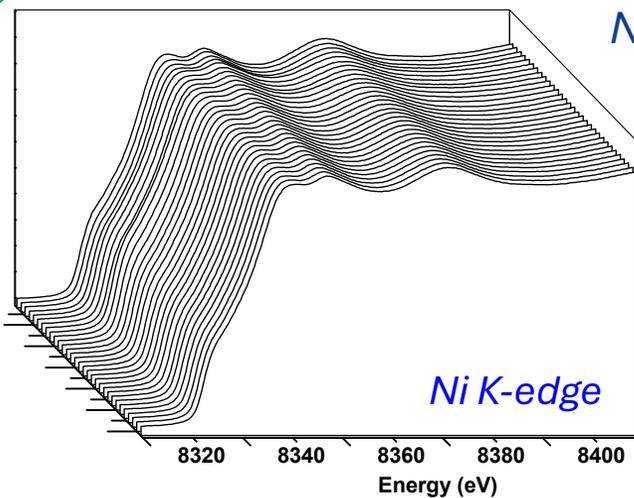


Same behaviour in 2 different cells!

ESR reaction: 500°C, 40 mL/min of He/EtOH/H₂O

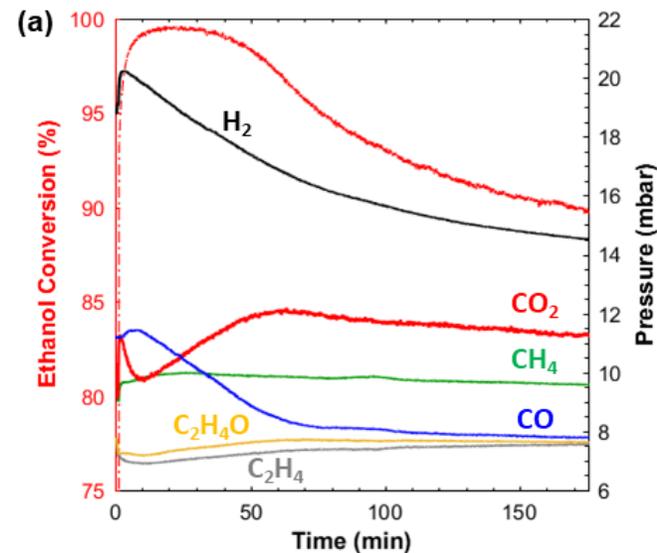


Operando



No evolution of the catalyst → Nothing happens?

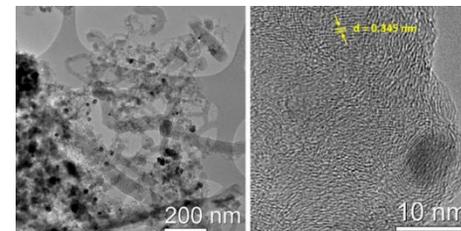
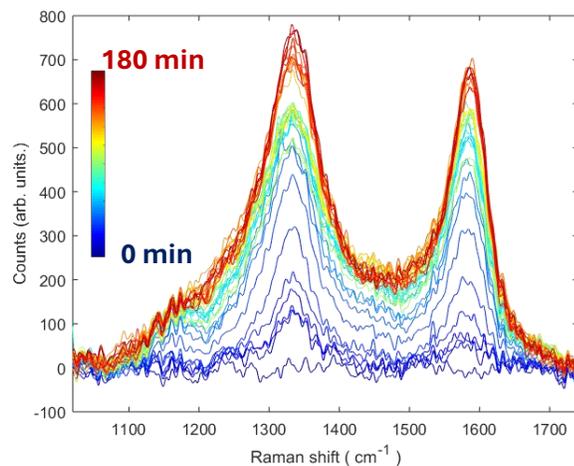
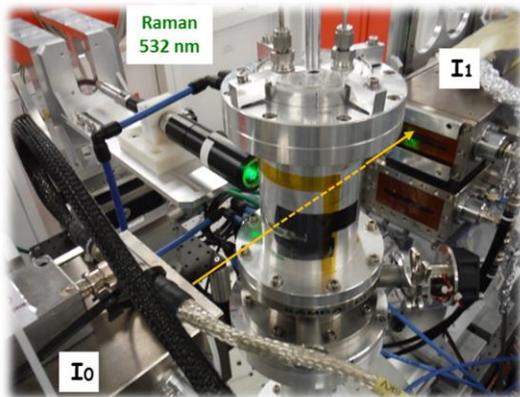
Mass Spectrometry



Deactivation of the catalyst! → Why?

Coke formation at the surface of the nanoparticles!

Coupling technique



I. Examples of *in situ/operando* cells

II. Multimodal characterisations and data analysis methodology

III. Scientific case: LDH used for Ethanol Stream Reforming

IV. Conclusions

- ✓ **What is your study case?**
 - Catalysis? Thermal, photo, electro?
 - Battery?
 - Liquid reaction?

- ✓ **What are the reaction conditions?**
 - Temperature, pressure,...

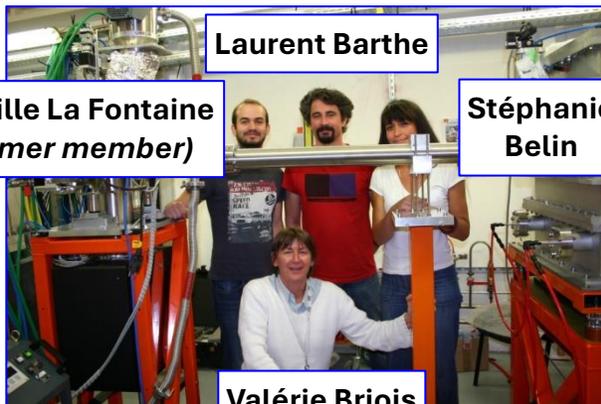
- ✓ **Do you want to perform coupling techniques?**
 - Raman, UV-visible, XRD,...

 **Choose/design the better *in situ/operando* cell**

- Adapt the cell to your experiments (windows type, thickness, ...)
- Optimize the preparation of the material (thickness, dilution with *ad hoc* dilutant if needed,...)
- Test the cell before *in situ/operando* measurements to be sure that activity is the same as at the laboratory



**INSTITUT DE CHIMIE
DE CLERMONT-FERRAND**



Laurent Barthe

**Camille La Fontaine
(former member)**

**Stéphanie
Belin**

Valérie Briois



lunesp



**Antonella Iadecola
(RS2E)**



Olga Roudenko



**Laboratoire
Catalyse & Spectrochimie**



A photograph of a modern building with a curved facade, featuring large glass windows and a prominent metal grid structure on the roof. The building is set against a clear blue sky. A yellow vertical bar is on the left side of the image, and a yellow sun icon is in the bottom left corner.

Thank you for your attention